

The Meudon PDR code

J. Le Bourlot
F. Le Petit
E. Roueff



- ✧ Introduction
- ✧ Benchmark exercise
- ✧ Photoprocesses and UV radiative transfer
 - exact / approximate
 - diagnostic tool?
- ✧ Chemical processes
 - H_2 formation
- ✧ Conclusions

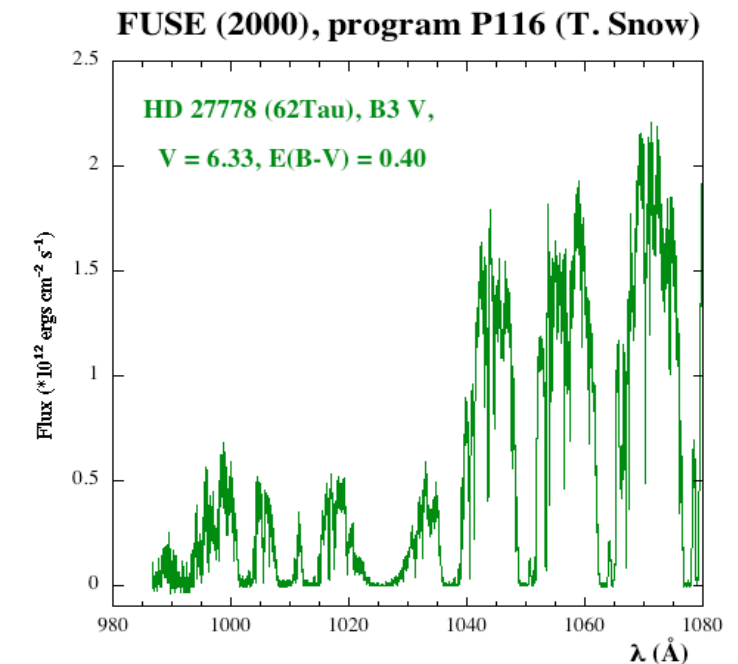
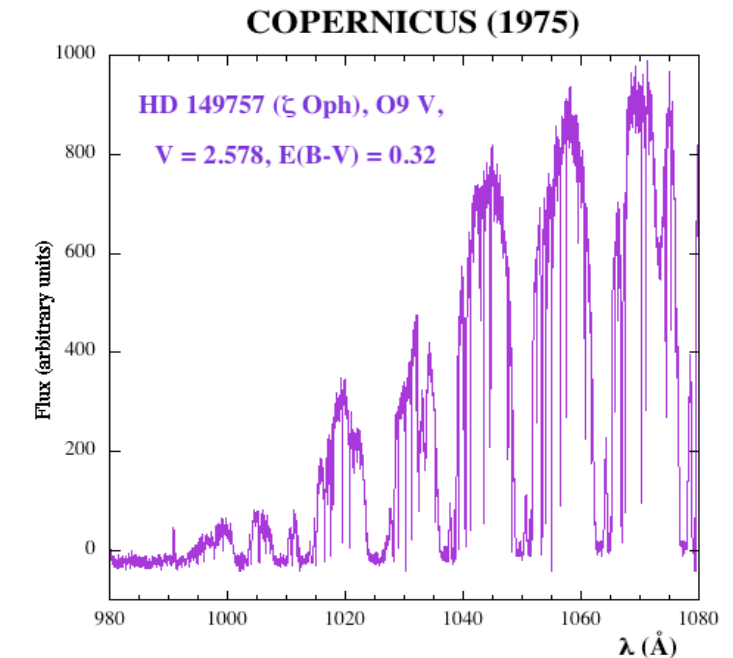
PDR models

Photon Dominated Regions :

- Diffuse and translucent clouds
- Edge of molecular clouds
- Damped Lyman α systems
- Circumstellar disks
- ...
- Dark clouds

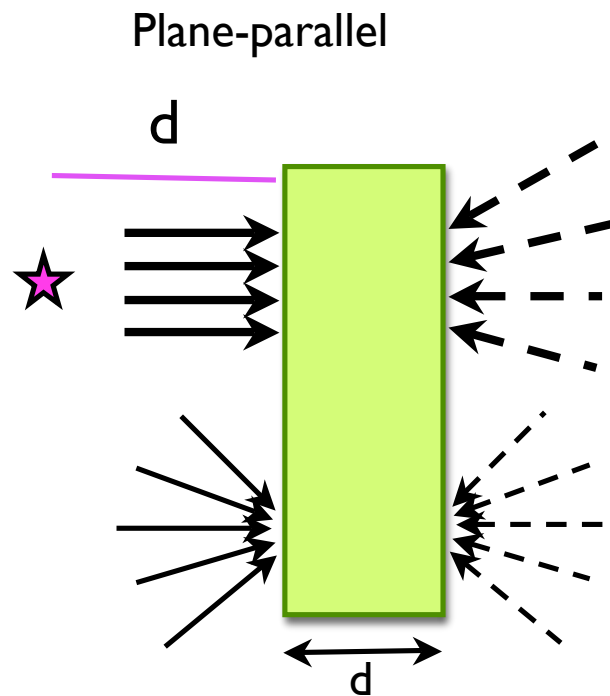
Some historical facts

- 70's : first detection of H_2 with Copernicus toward bright stars : Black, Dalgarno, Glassgold, Hollenbach, Jura,
Stationary models - H/H_2 transition-
- 80's : GHRS of HST UV spectra longwards 120 nm C^+ / C / CO transition : van Dishoeck, Black, Sternberg, Viala, Flower, Pineau des Forêts, Le Bourlot, Roueff ...
- 90's : ISO H_2 and atomic fine structure line seen in emission !
- 2000's : FUSE : fainter sources
- present and future : Spitzer, Herschel, ALMA, JWST, ... ?



PDR models

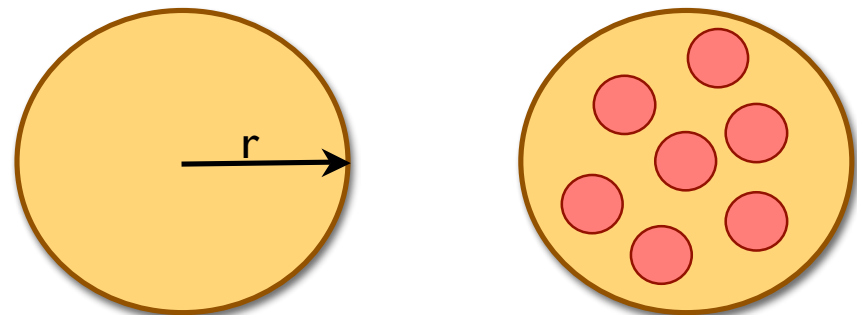
Geometry



- 1 or 2 side illumination
- Isotropic or perpendicular



Spherical



- Isotropic illumination
- $r \longrightarrow \infty$ to mimic a plane parallel case

PDR models

Equation of
state (n , p), $A_{\text{v,tot}}$

U_{inc} , ξ , grain properties,
elemental abundances

input

UV

UV

Abundances
Excitation & Emissivities
Gas and dust temperatures

output

Column
densities

Intensities /
Spectra

Problem to be solved :

Compute all local quantities :

- abundances
- populations
- local emissivities
- Heating/cooling processes
- temperatures : gas and dust particles

UV radiative transfer

Chemistry

Thermal balance

Benchmark of PDR codes (Leiden)

(Röllig et al. - A&A 467, 187, 2007)

Aikawa	Lee, Herbst, Pineau des Forêts, Le Boulrot, Aikawa	Analytical formulae for H ₂ and CO photodissociation
Lee96mod	Lee, Herbst, Pineau des Forêts, Roueff, Le Boulrot	Analytical formulae for H ₂ and CO photodissociation
Kosma	Störzer, Köster, Zilinsky, Jeyakumar, Röllig	spherical geometry
Bensch	Störzer, Köster, Zilinsky, Jeyakumar, Bensch	spherical geometry
HTBKW	Hollenbach, Tielens, Burton, Kaufman, Wolfire	Simplified H ₂
Leiden	Black, van Dishoeck, Jansen, Jonkheid	Detailed physics
Cloudy	Ferland, van Hoof, Abel, Shaw	Detailed physics
Meudon	Le Boulrot, Roueff, Le Petit	Detailed physics
Sternberg	Sternberg, Dalgarno	Detailed physics
Meijerink	Meijerink, Spaans	XDR
Costar	Kamp, Bertoldi, van Zadelhoff	Circumstellar disks
UCL	Viti, Thi, Bell	Time dependence

Benchmark exercise

Simplification

Chemistry based on : H, He, C, O

31 species

Chemistry : UMIST99

H₂ formation rate fixed : $R = 3 \times 10^{-18} T^{1/2} \text{ cm}^3 \text{ s}^{-1}$ (2×10^{-17} at 50 K)

2 types of models

F1 $n = 10^3 \text{ cm}^{-3}, \chi = 10$	F2 $n = 10^3 \text{ cm}^{-3}, \chi = 10^5$	T fixed 50 K	Test of photo-processes
F3 $n = 10^{5.5} \text{ cm}^{-3}, \chi = 10$	F4 $n = 10^{5.5} \text{ cm}^{-3}, \chi = 10^5$		
V1 $n = 10^3 \text{ cm}^{-3}, \chi = 10$	V2 $n = 10^3 \text{ cm}^{-3}, \chi = 10^5$	T variable	Test of thermal balance
V3 $n = 10^{5.5} \text{ cm}^{-3}, \chi = 10$	V4 $n = 10^3 \text{ cm}^{-3}, \chi = 10^5$		

Benchmark exercise

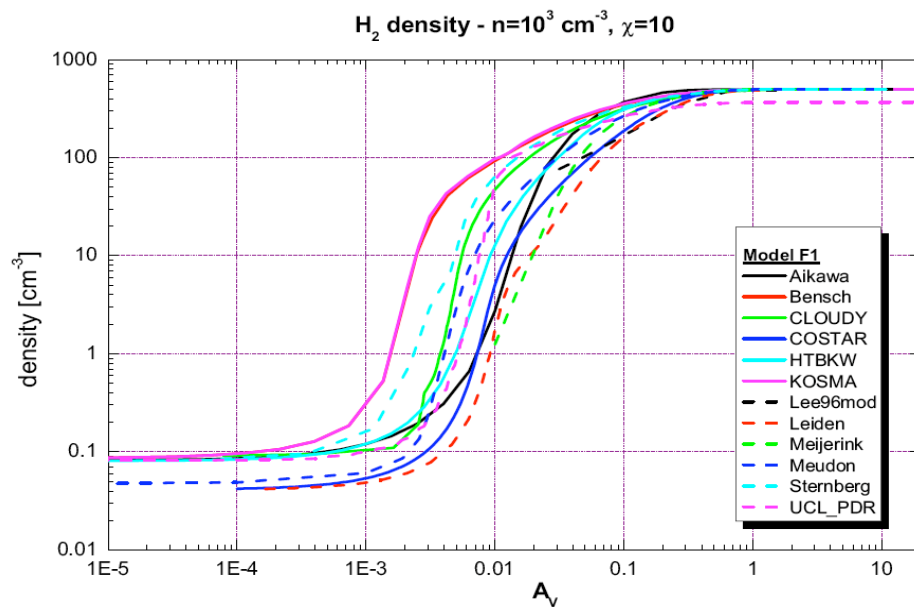
FI model

$\chi = 10$

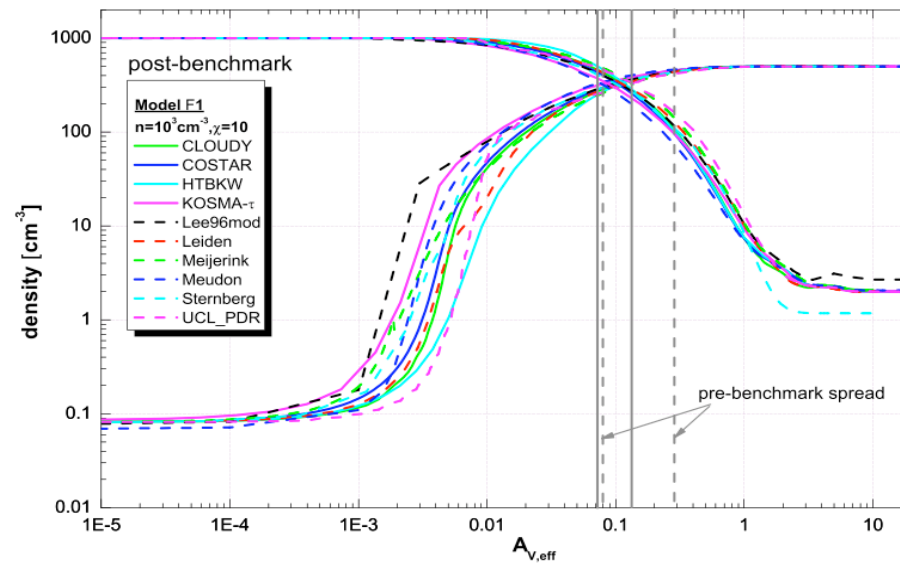
$n_H = 1000 \text{ cm}^{-3}$

$T = 50 \text{ K}$

Before comparison



After comparison



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Characteristics

- Plane parallel & steady state
- Detailed UV radiative transfer for photo-processes
 - with discrete gas absorption in H, H₂, HD, CO, ... (~35000 transitions)
 - continuous absorption from dust particles
- Chemistry
- Thermal balance
- Detailed balance of molecular populations in H₂, CO, CS, H₂O, HCO⁺, H₃⁺, ...

<http://aristote.obspm.fr/MIS>

Various improvements since benchmark exercise

- exact radiative transfer (Goicoechea & Le Boulrot, AA 467, 1, 2007)
- infrared pumping of H₂O by dust radiation (Gonzalez-Garcia et al. submitted)
- H₂ formation with moment equations (work in progress with O. Biham)
-

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Inversion tool

- analyse lines
- derive physical conditions (T , n_H , ...)
- Hint at evolution paths
- Caveat : Reality is too sophisticated

Experiment *in silico*

- Test physical hypotheses
- Assert importance of physical parameters
- Suggest new paradigms
- Caveat : Reality is too sophisticated

We must adapt our limited tools and computing power to our goals

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Radiative transfer

Decoupling between UV and IR / radio

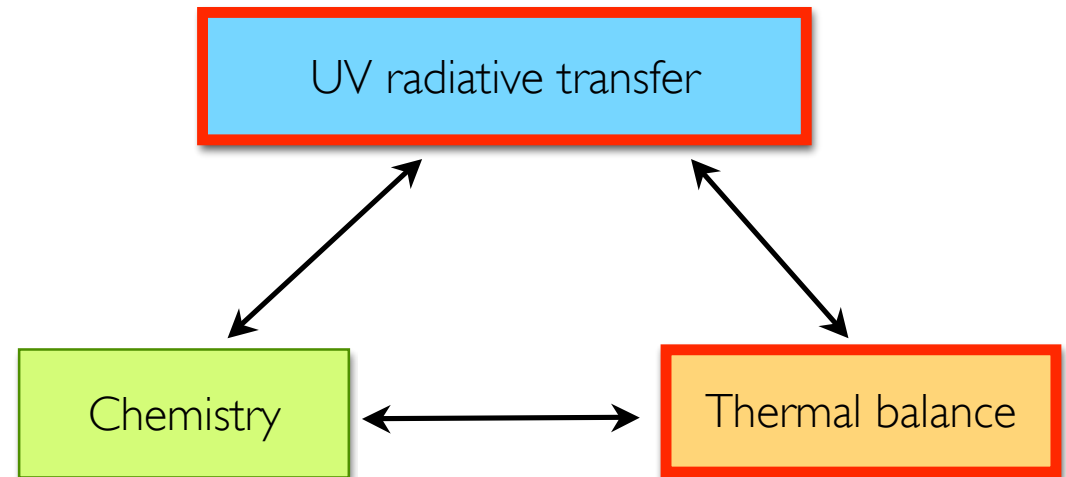
UV radiative transfer

- dependence on the excitation status (H_2 , HD, CO)
- dependence on the grain properties
- absorption
- angular redistribution
- different levels of approximations
 - ✧ “FGK” (Federman et al. 1979, ApJ 227, 466) self shielding approximation
 - ✧ “exact”

• IR / radio radiative transfer :

Statistical equilibrium equations (collisional + radiative excitation and de-excitation)

- [C II] 158 μm and [Si II] 35 μm (Barinovs & van Hemert, 2005, ApJ 620, 537)
- [O I] 63, 145 μm (Abrahamson et al. 2007, ApJ 654, 1171)



- [C I] 370, 610 μm (Abrahamson 2007)
- CO [H_2 : Flower 2001, H revisited by Shepler et al, 2007)
- H_2O [H_2 : Green, Phillips, revisited by Dubernet, Grosjean et al.)
- H_2 : collisions with H, revisited by Wrathmall and Flower, 2006; non-reactive only

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One can not avoid the transfer equation

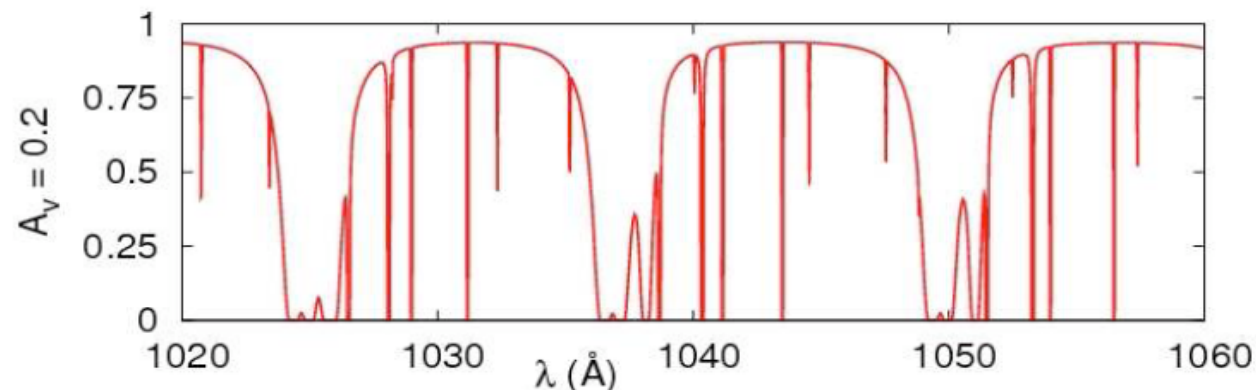
$$\mu \frac{\partial I_\lambda(\tau_\lambda, \mu)}{\partial \tau_\lambda} = I_\lambda(\tau_\lambda, \mu) - S_\lambda(\tau_\lambda) - \frac{\omega_\lambda(\tau_\lambda)}{2} \int_{-1}^{+1} p_\lambda(\mu, \mu') I_\lambda(\tau_\lambda, \mu') d\mu'$$

with
$$\omega_\lambda(\tau_\lambda) = \frac{\sigma_\lambda^D(\tau_\lambda)}{\kappa_\lambda^G(\tau_\lambda) + \kappa_\lambda^D(\tau_\lambda) + \sigma_\lambda^D(\tau_\lambda)}$$

I_λ : specific intensity
 S_λ : Source function
 ω_λ : effective albedo
 p_λ : angular redistribution function

Solution through spherical harmonics

- First proposed by Flannery et al., ApJ 236, 598, 1980
- Extended by Roberge, ApJ 275, 292, 1983 to include embedded sources
- Extended by Goicoechea & Le Boulot (2007) with variable coefficients including discrete transitions



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Photo-reactions probability calculations

$$k = \int \sigma(\lambda) I(\lambda) d\lambda$$

$\sigma(\lambda)$: photodissociation cross section cm^2

$I(\lambda)$: mean intensity of the radiation in photons $\text{cm}^{-2} \text{s}^{-1} \text{\AA}^{-1}$

In presence of resonances :

$$k = \frac{\pi e^2}{mc^2} \lambda_u^2 f_u \eta_u I_u$$

8.85×10^{-21}

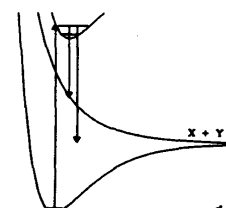
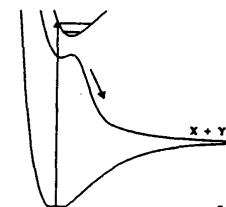
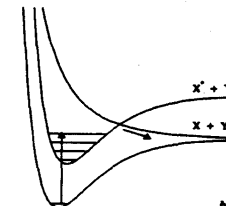
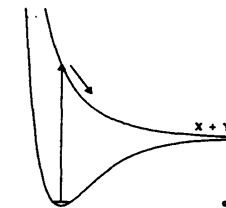
Frequent approach : $k = k_0 \cdot \chi \cdot \exp(-\beta A_V)$

Example : $\text{S} + h\nu \rightarrow \text{S}^+ + e^-$ $k = 5.9 \times 10^{-10} e^{-2.58 A_V}$

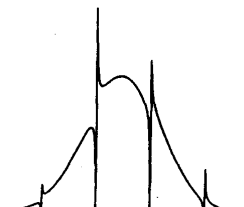
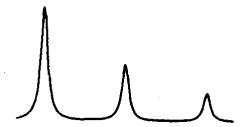
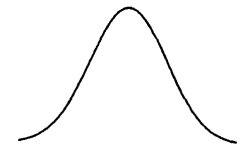
- All possible channels should be considered
- dependence on the grain properties and intensity of the radiation field (values different for “Mathis” and “Draine” radiation fields)

cf Ewine’s talk

Photodissociation mechanism



Cross section



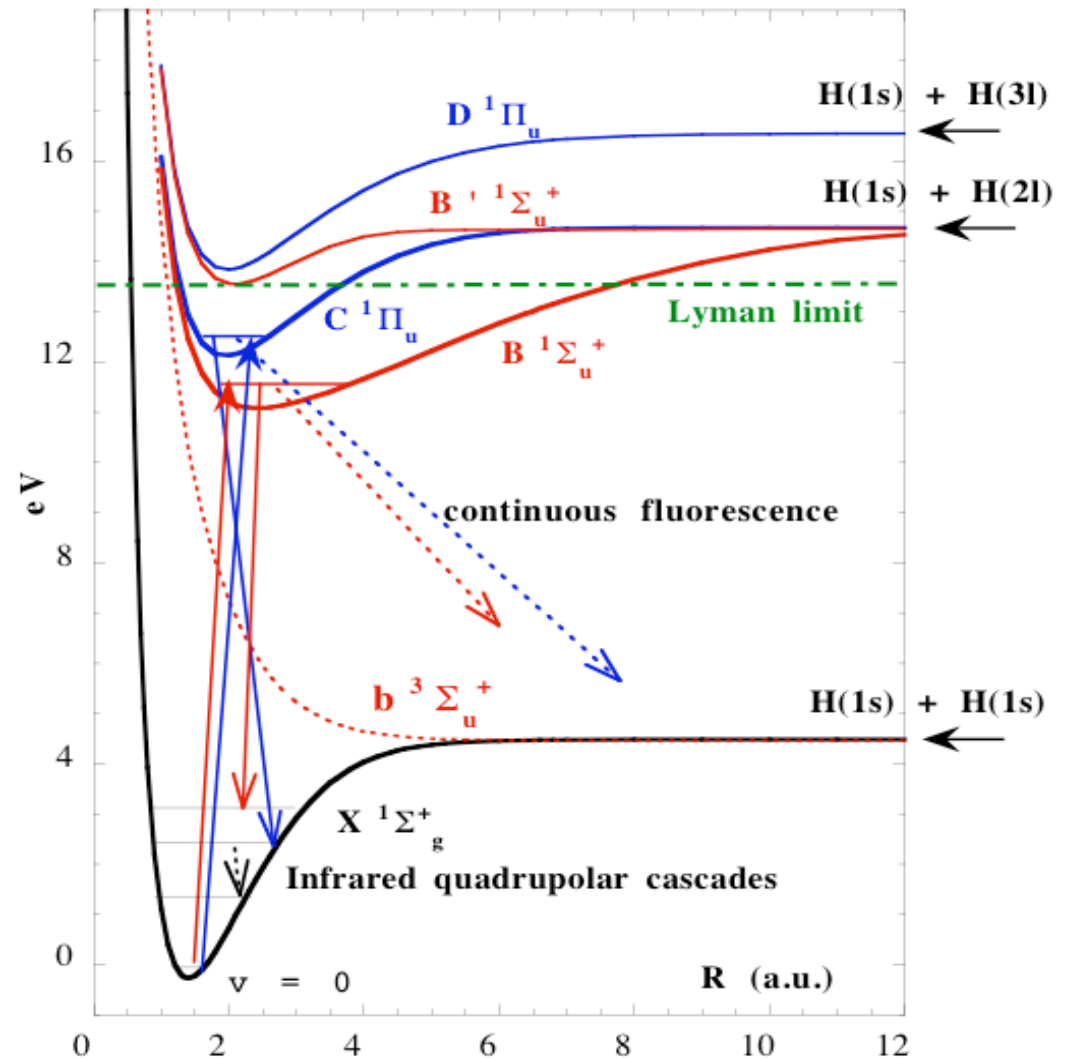
From van Dishoeck (1988)

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H₂ photodissociation probability

Dependence on

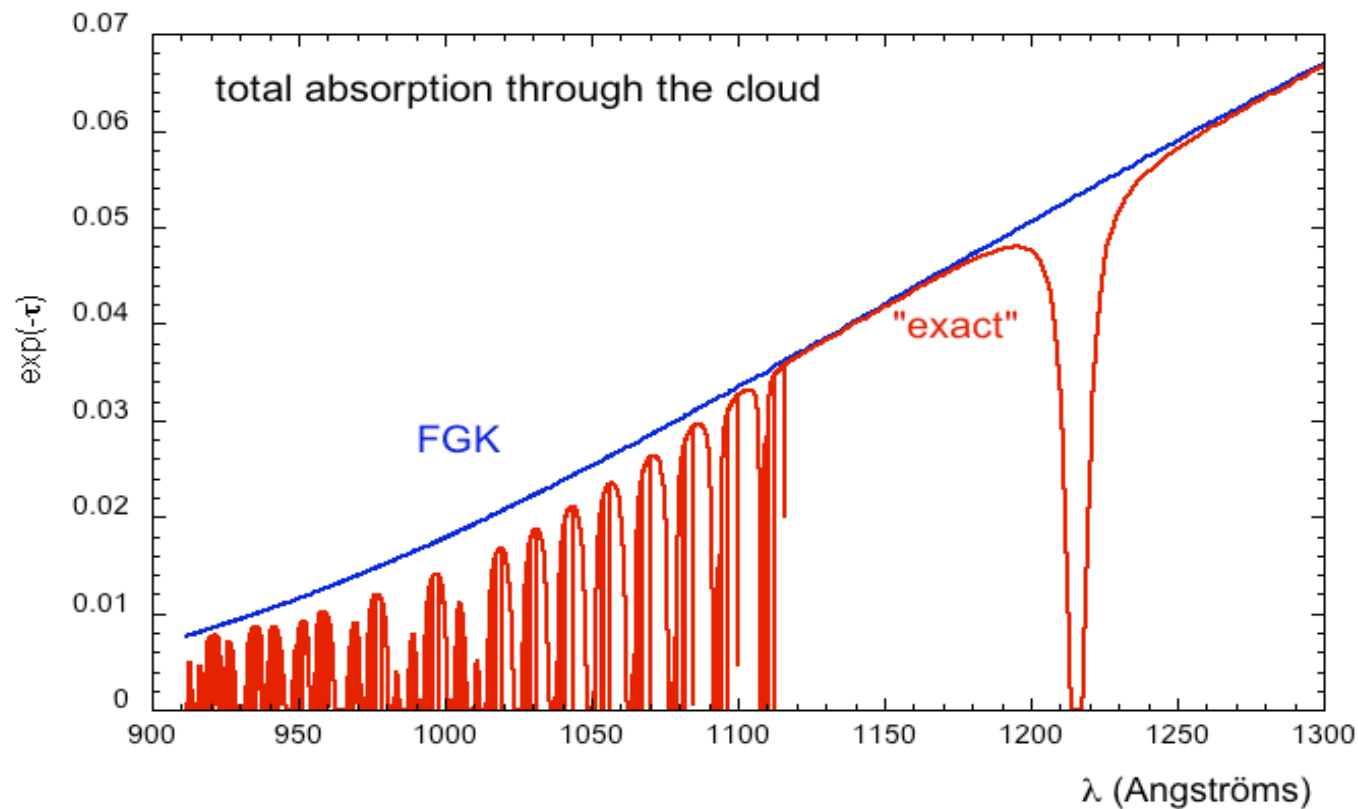
- incident radiation
- H₂ excitation
- molecular physics data
 - J dependent oscillator strengths and photodissociation probabilities (Abgrall & Roueff)
data available at <http://molat.obspm.fr>
 - (de)excitation collision rates by H, He, H₂, electrons, H⁺, ...
(new calculations by Wrathmall & Flower in J. Phys B 39, L249 (2006) for H₂ + H)
data available at http://massey.dur.ac.uk/saw/webseite_data



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FGK versus “exact” UV transfer

- ✧ 2-side illumination model with $\chi = 1$, $n_H = 100 \text{ cm}^{-3}$
- ✧ exact radiative transfer for H and H_2 absorption up to $J = 4$

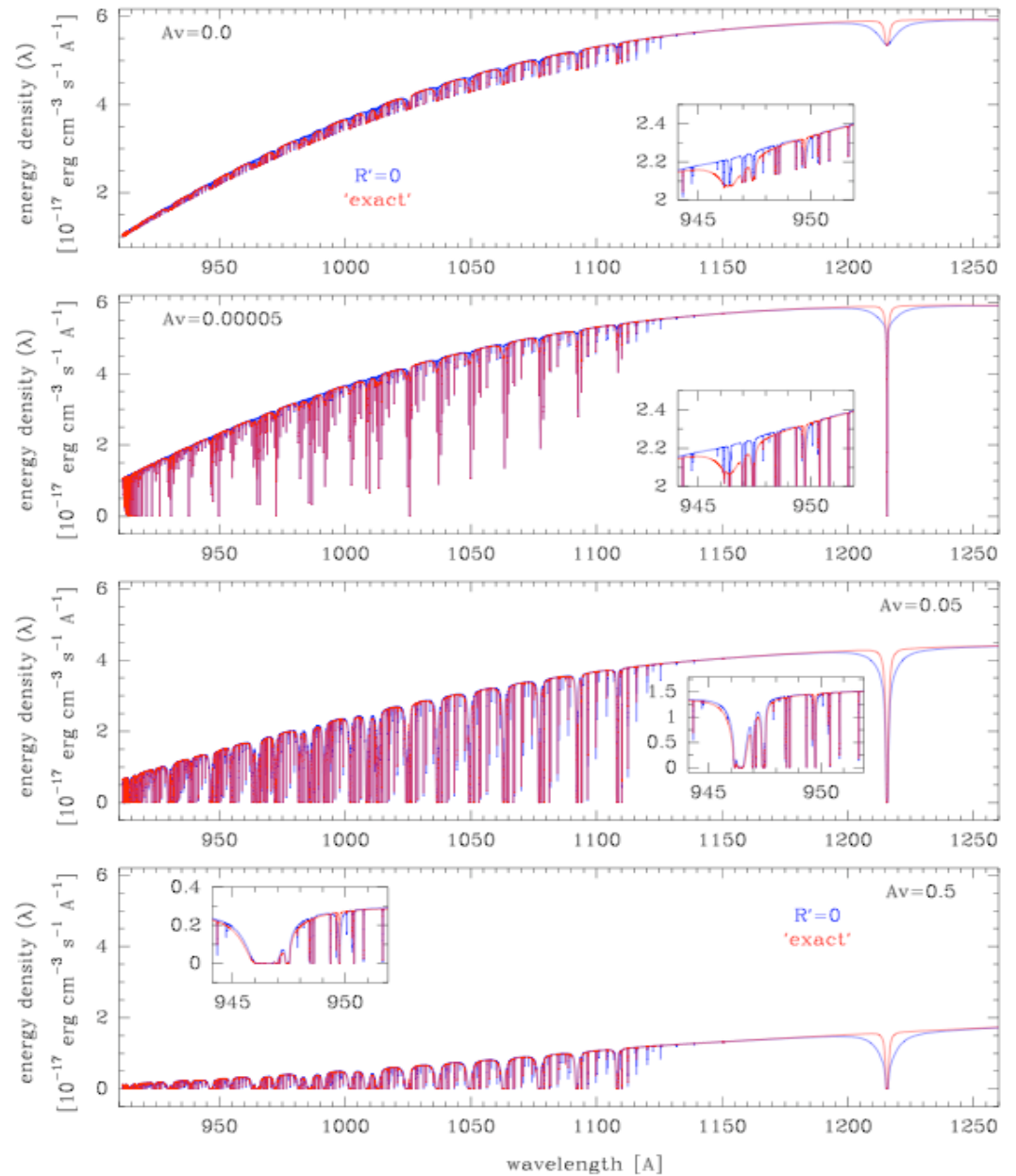
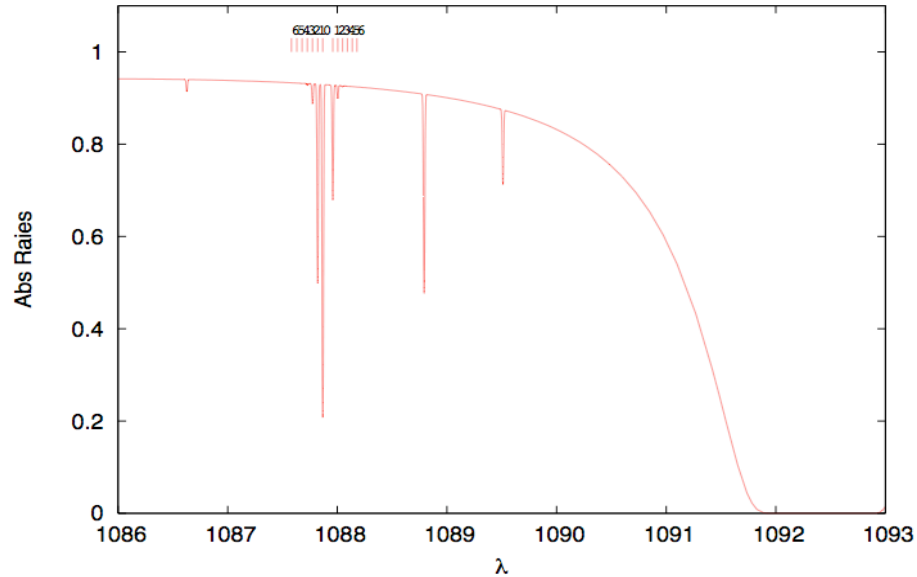


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Exact Transfer :

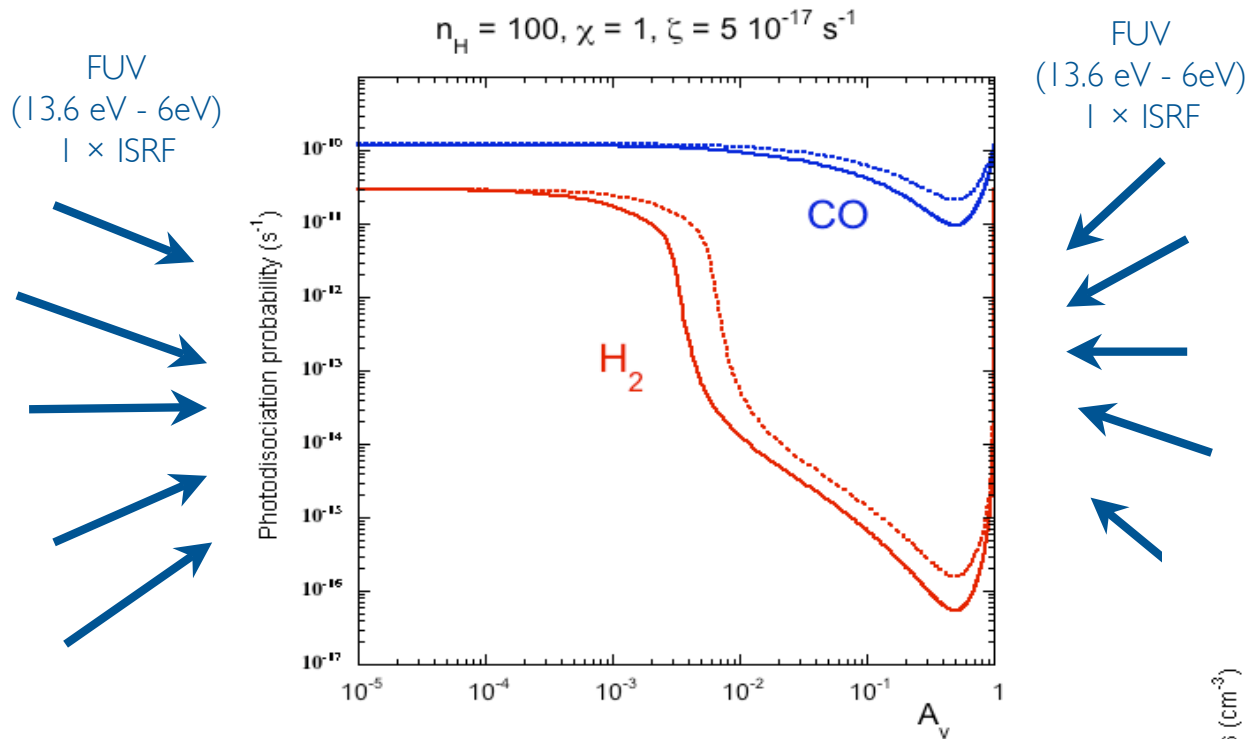
Goiecoechea & Le Boulrot (AA 467, I, 2007)

CO transitions:



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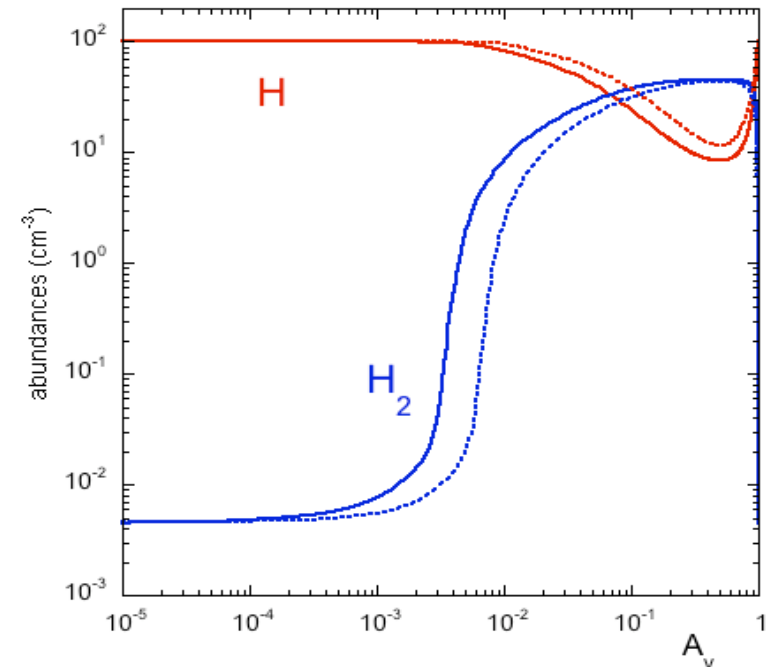
FGK (dotted lines) versus “exact” (full lines) UV transfer



Included cross-sections :

H_2 : PD
 CO : PD
 C : PI
 S : PI
 CH : PD + PI
 C_2 : PD + PI
 C_3 : PD + PI
 CN : PD
 OH : PD + PI
 H_2O : PD + PI
 NH : PD
 HF : PD

- ✧ Decrease of photodissociation probabilities when exact transfer is included (role of overlap with H_2 transitions)
- ✧ H/ H_2 transition towards smaller A_v
- ✧ same trend for all photodissociation rates



Meudon PDR code

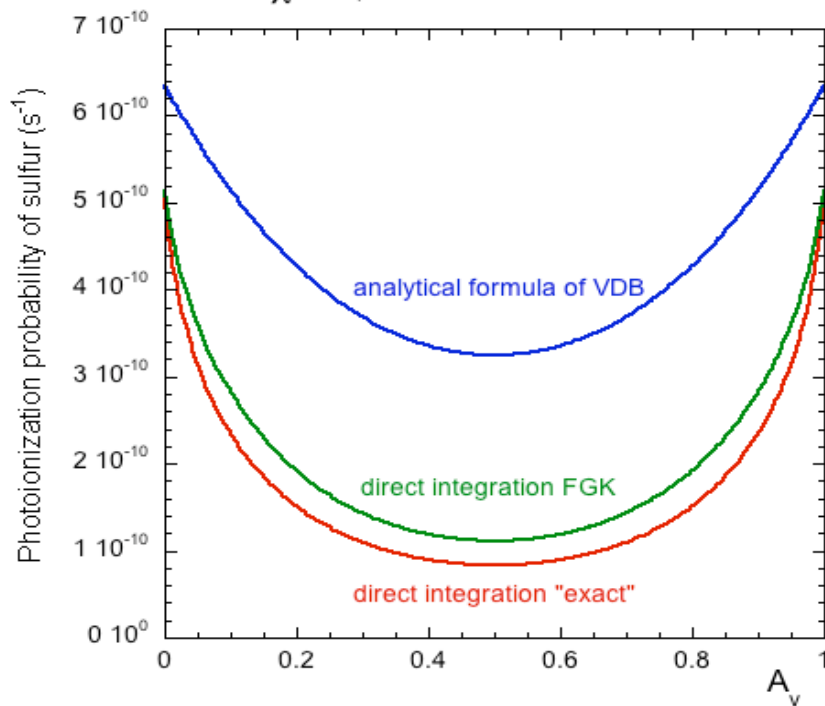
Exact transfer allows to know the radiation field at each point;

Photodissociation probability

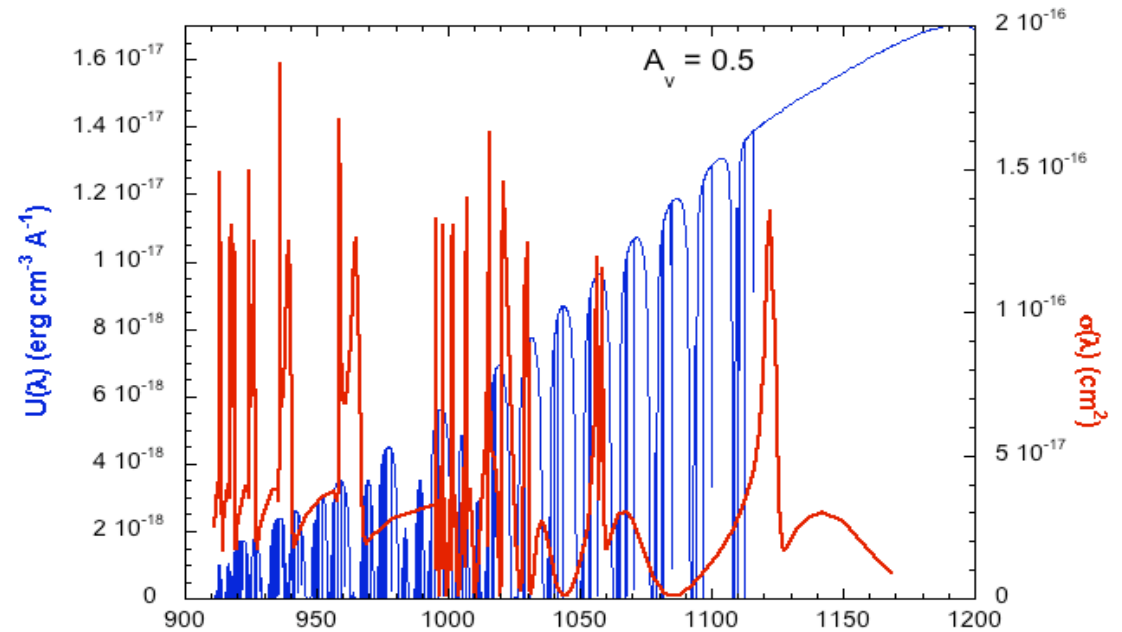
$$k = \int \sigma(\lambda) I(\lambda) d\lambda$$

Sulfur example :

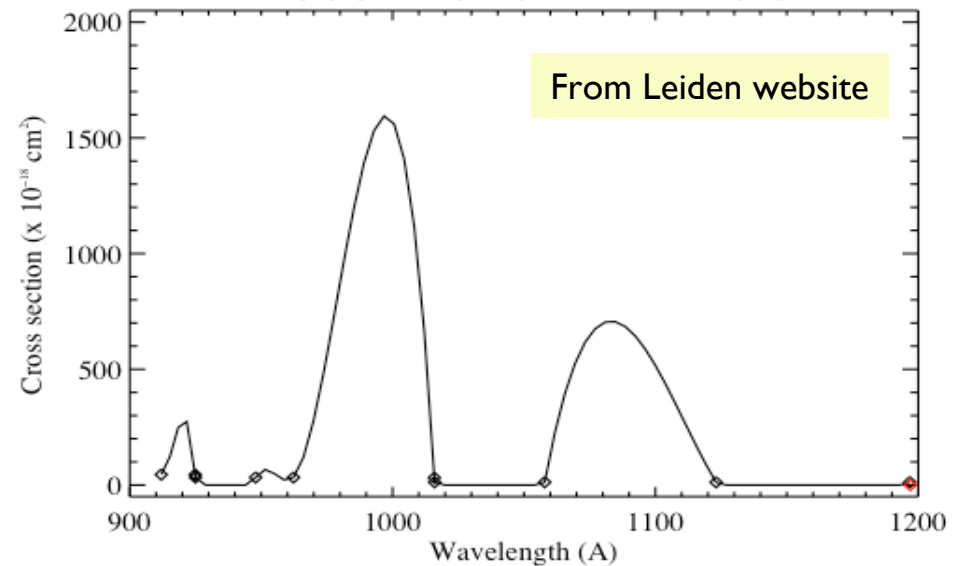
$n_H = 100 \text{ cm}^{-3}$, $\zeta = 5 \cdot 10^{-17} \text{ s}^{-1}$,
 $\chi = 1$, 2-side illumination



σ from Topbase

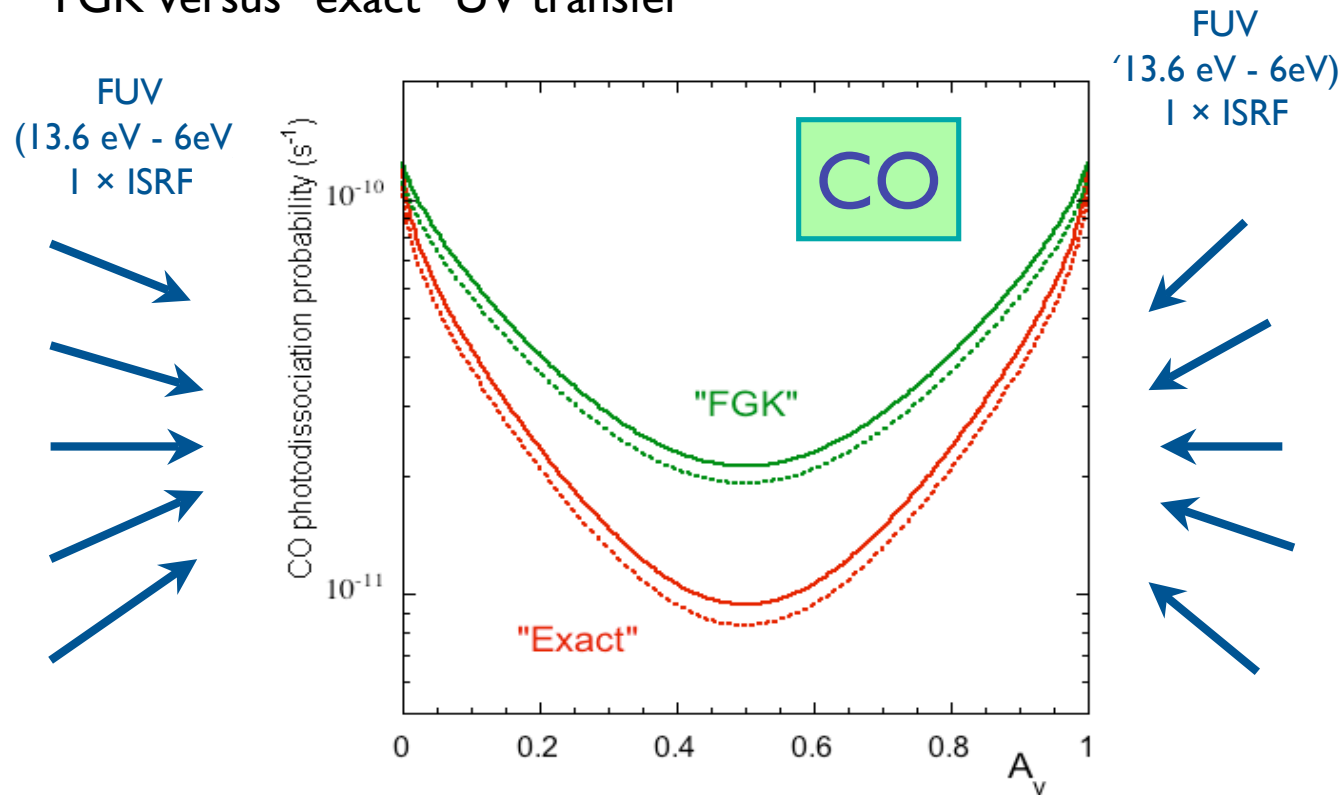


Photoionization continuum data of S



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FGK versus “exact” UV transfer



☛ full lines: with revised oscillator strengths of Eidelsberg et al. 2006, ApJ 647, 1543, for E-X, B-X and W-X transitions

Modelled column densities

species	exact	FGK
H	3.6(20)	5.0(20)
H ₂	7.6(20)	6.9(20)
C ⁺	1.5(17)	1.5(17)
C	1.3(15)	8.1(14)
S	3.2(13)	2.7(13)
CO	1.2(13)	6.5(12)
CH	1.5(13)	1.2(13)
CN	9.1(10)	4.0(10)
OH	4.3(12)	3.8(12)
H ₃ ⁺	1.4(13)	1.2(13)

E. Bayet et al., in preparation

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Example of diffuse clouds

2-side model : $\chi = 1$

2 cases : - FGK

- "exact" transfer up to $J = 3$

"exact" transfer required for diffuse regions

n_H (cm ⁻³)		N(H)	N(H ₂)	f	T _{moy}	T ₀₁
10	FGK	1.3E20	7.3E15	1.1(-4)	695	-2080
	exact	1.3E20	6.2E16	9.5(-4)	655	321
30	FGK	1.2E20	4.2E18	6.5(-2)	221	151
	exact	1.1E20	9.8E18	0.15	192	142

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Thermal balance

Several uncertainties

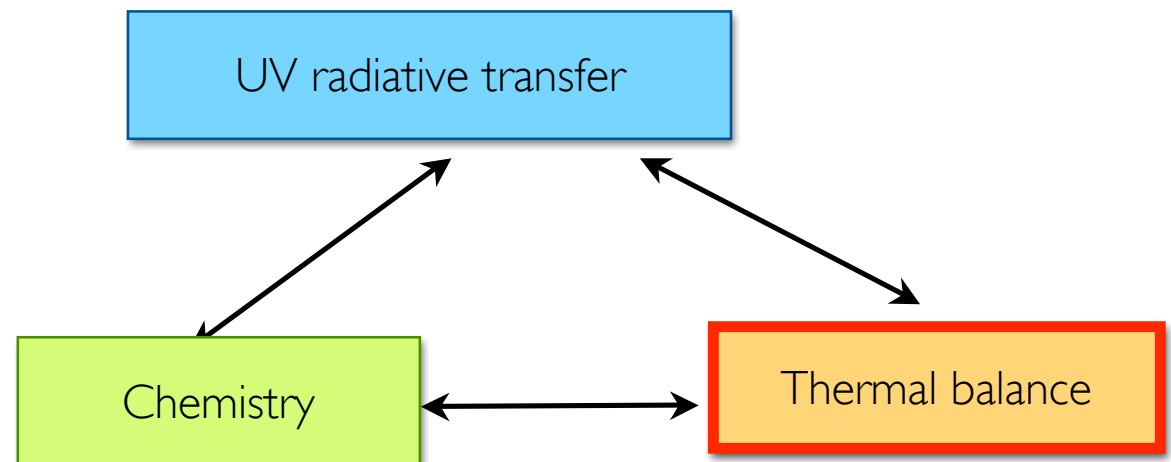
Heating :

- Photo-electric effect on grains
- H_2 : vibrational de-excitation, dissociation, formation
- Cosmic ray ionization; secondary UV photons
- Gas/grain interaction (if $T_{\text{grain}} > T_{\text{gas}}$)
- chemical heating

Cooling :

- [C II] 158 μm
- [O I] 63, 145 μm
- [C I] 370, 610 μm
- H_2 , CO, H_2O , HCO^+ , CS, OH, ...
- Gas/grain interaction (if $T_{\text{grain}} < T_{\text{gas}}$)

Escape probabilities



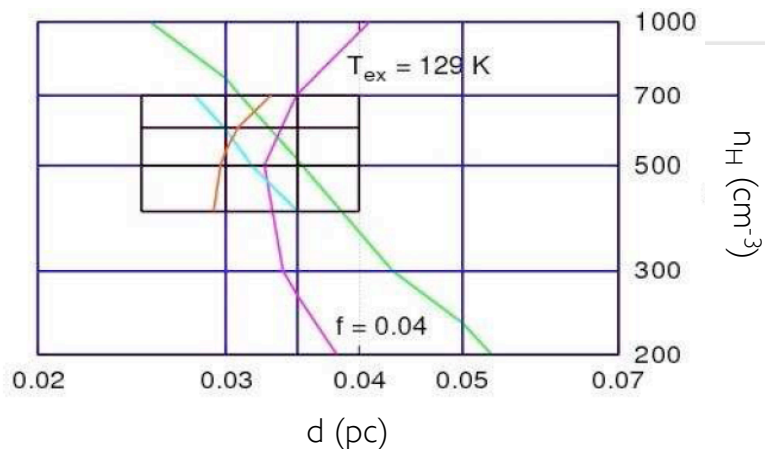
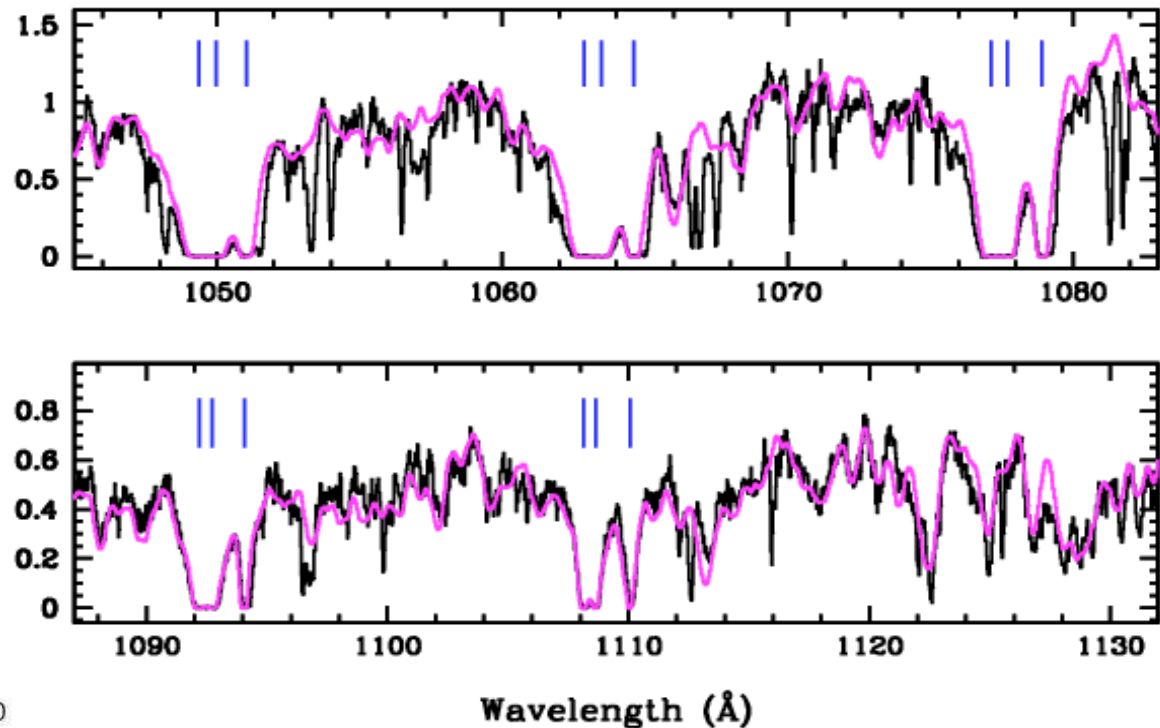
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Observation of circumstellar H₂ towards HD 76534

Martin-Zaïdi et al.
submitted

Observational information

- ✧ f molecular fraction
- ✧ excitation temperature of lowest H₂ levels



Space parameters

- ✧ n_{H}
- ✧ Distance of the star to the envelope

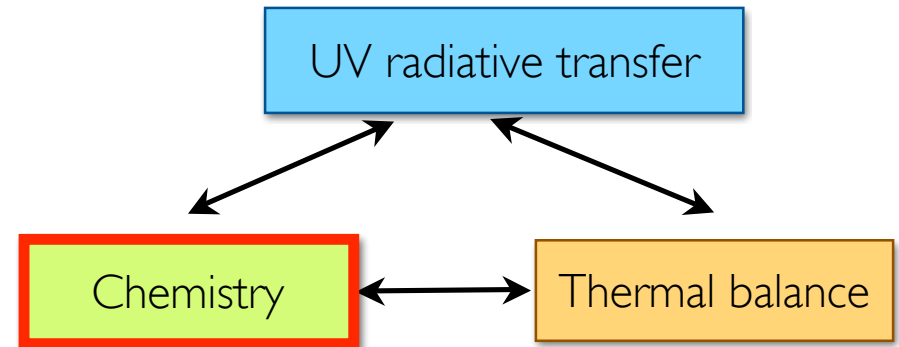
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Chemistry

Typically 10 more reactions than involved species

All binary reactions relevant to dark cloud chemistry (cf Eric's and Tom's talk) + :

- Photo-reactions
- Formation reaction of H_2 on grains
- Reactions with atoms (H, N, O, C, S, ...) and ions (charge transfer included)
- Small endothermicities may be overcome at the edge of the PDRs ; role of reactions with vibrationally excited H_2



Free chemical reaction rates data bases

UMIST : <http://www.udfa.net/>

E. Herbst : <http://www.physics.ohio-state.edu/~eric/research.html>

M. Larsson : Dissociative recombination reactions : <http://mol.physto.se/moldata>

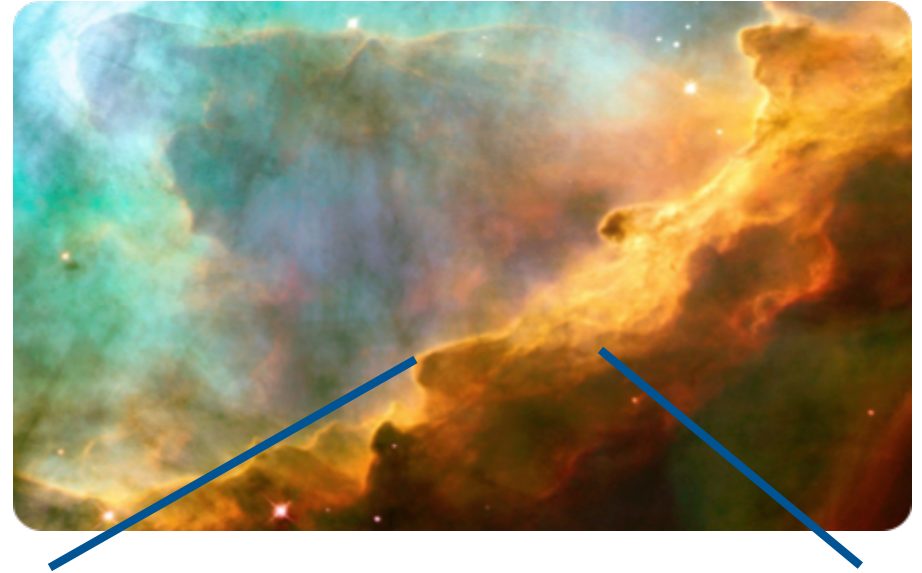
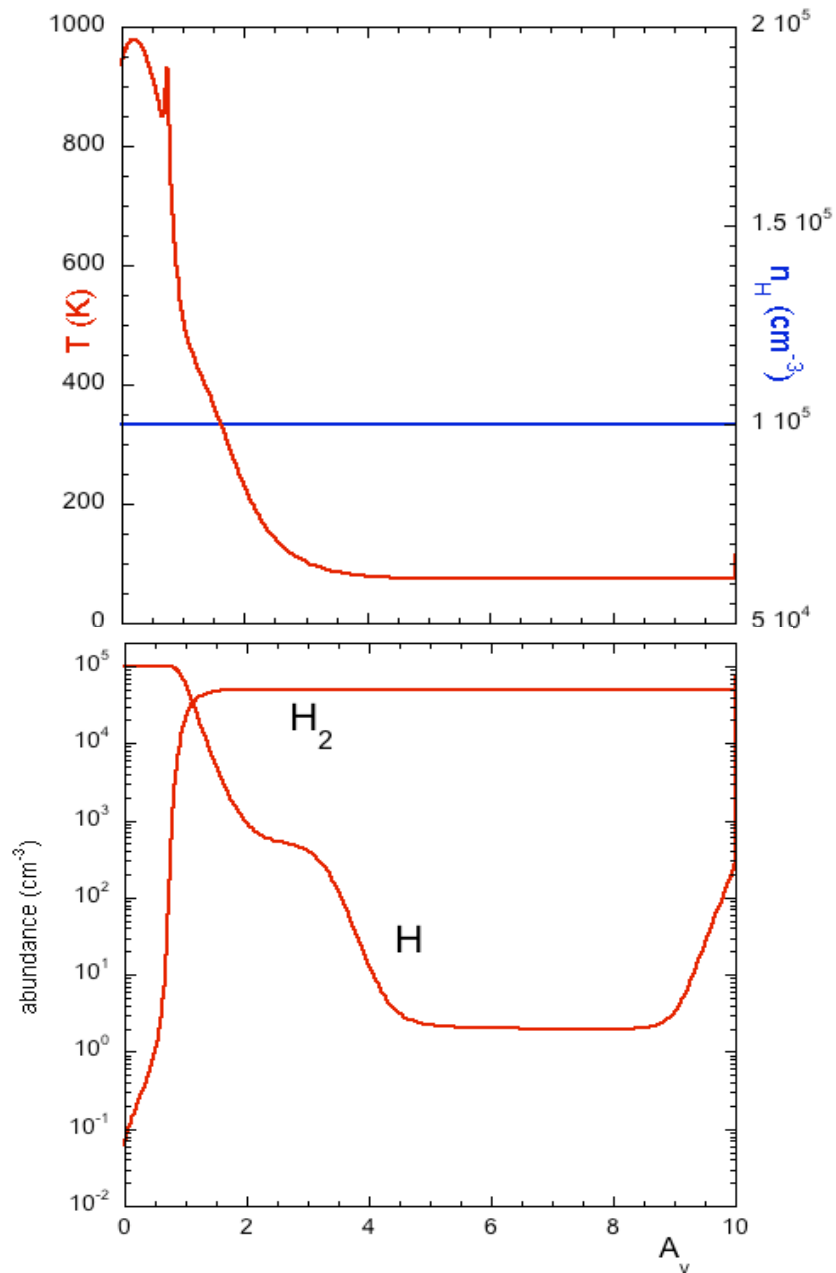
Photoprocesses : Leiden : <http://www.strw.leidenuniv.nl/~ewine/photo>

SWRI : <http://amop.space.swri.edu/>

Meudon : <http://aristote.obspm.fr/MIS>

KIDA : future

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Physical structure

$$n_{\text{H}} = 10^5 \text{ cm}^{-3}$$

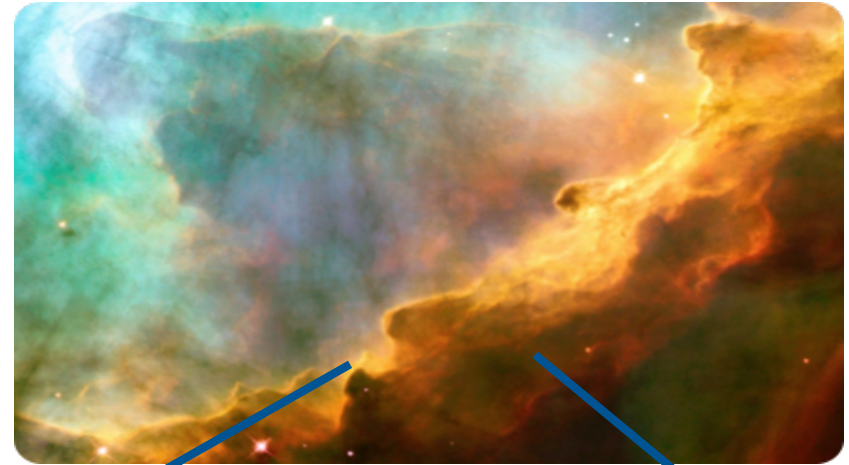
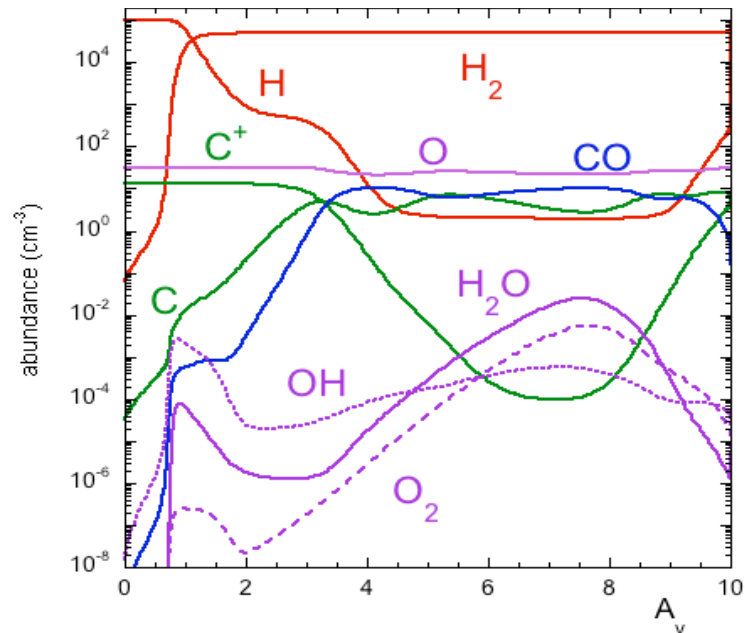
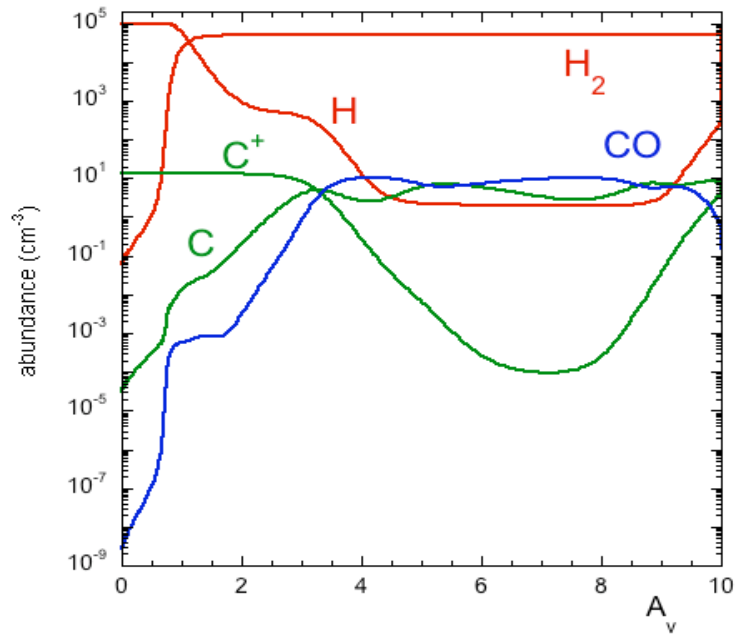
$$A_{\text{v,tot}} = 10 \approx d = 0.061 \text{ pc}$$

$$\chi_{\text{left}} = 10^5$$

$$\chi_{\text{right}} = 1$$

$$\xi = 5 \cdot 10^{-17} \text{ s}^{-1}$$

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$$n_H = 10^5 \text{ cm}^{-3}$$

$$A_{v,\text{tot}} = 10 \approx d = 0.061 \text{ pc}$$

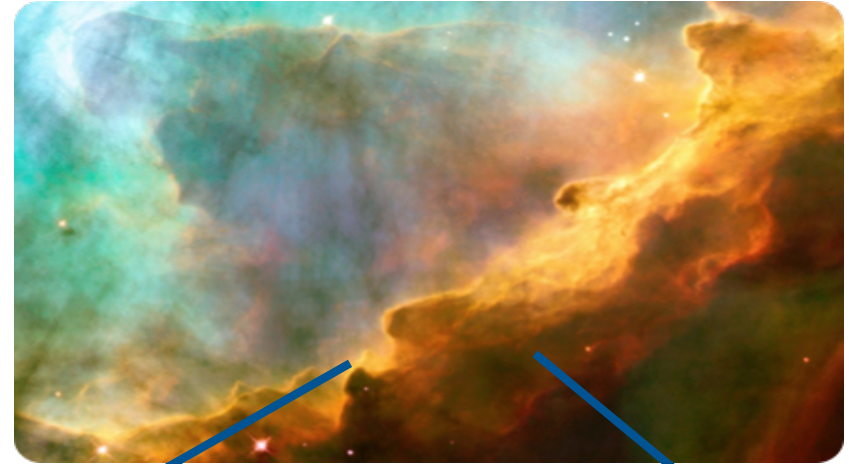
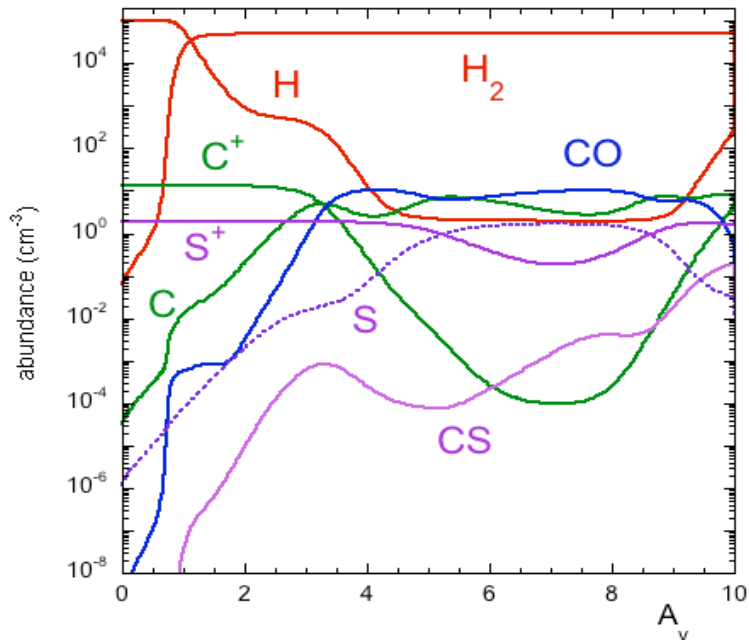
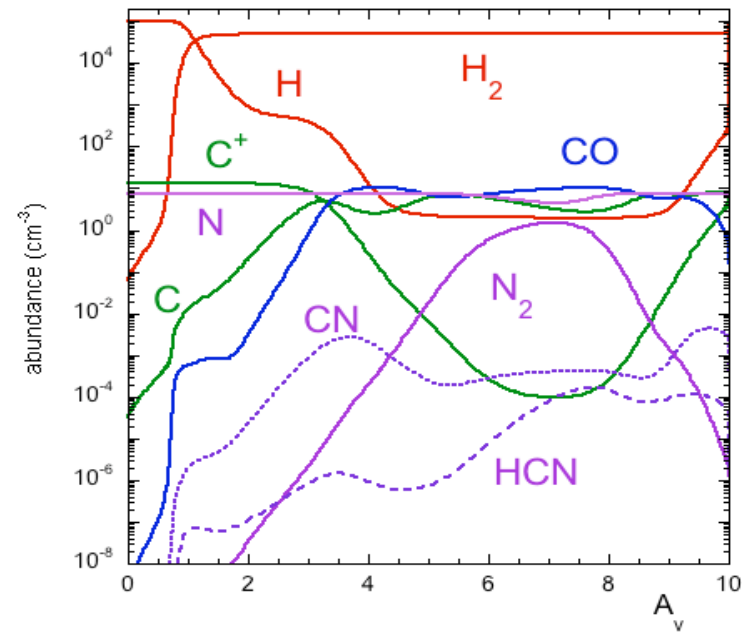
$$\chi_{\text{left}} = 10^5$$

$$\chi_{\text{right}} = 1$$

$$\xi = 5 \cdot 10^{-17} \text{ s}^{-1}$$

Carbon and oxygen chemistry

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$$\begin{aligned}
 n_H &= 10^5 \text{ cm}^{-3} \\
 A_{v,\text{tot}} &= 10 \approx d = 0.061 \text{ pc} \\
 \chi_{\text{left}} &= 10^5 \\
 \chi_{\text{right}} &= 1 \\
 \xi &= 5 \cdot 10^{-17} \text{ s}^{-1}
 \end{aligned}$$

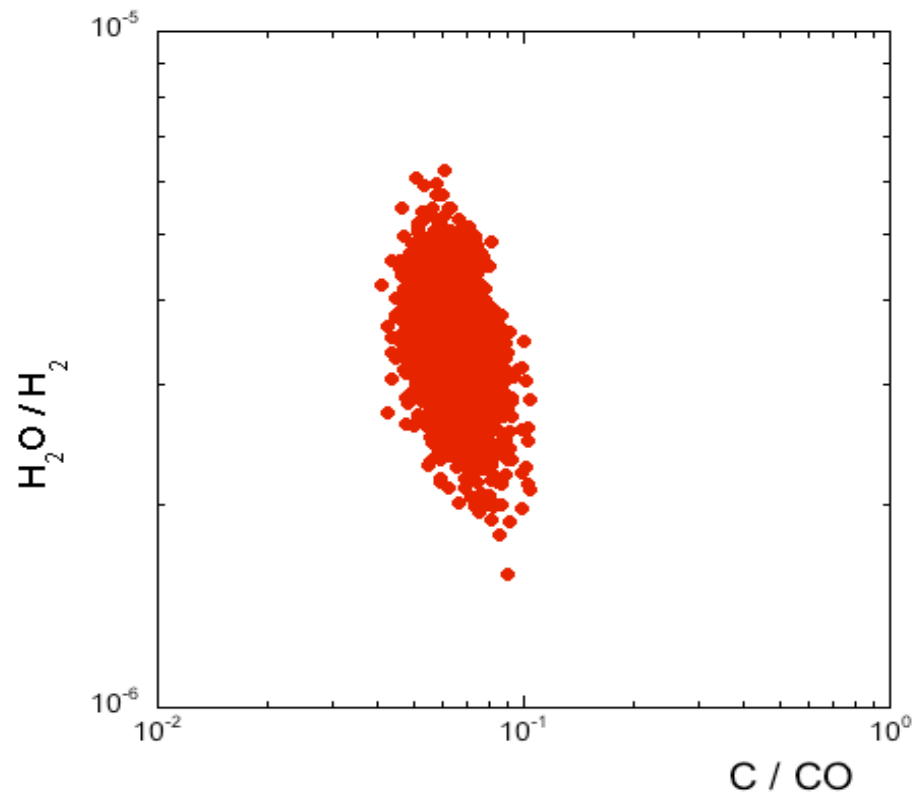
Nitrogen and sulfur chemistry

Chemical uncertainties

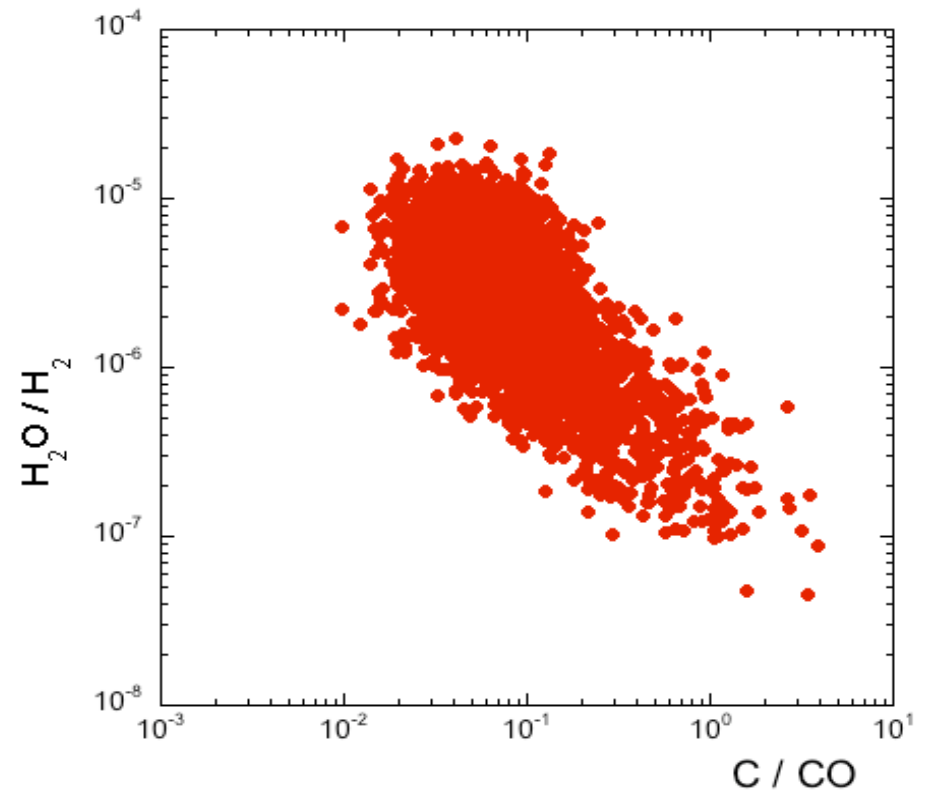
Basic question for models of any region

- Example : $n_H = 2 \cdot 10^3 \text{ cm}^{-3}$, $T = 10 \text{ K}$ et $\zeta = 5 \times 10^{-17} \text{ s}^{-1}$
- chemical network : 200 species - ≈ 2000 reactions

Random variation of reaction rate coefficients by a factor of 2



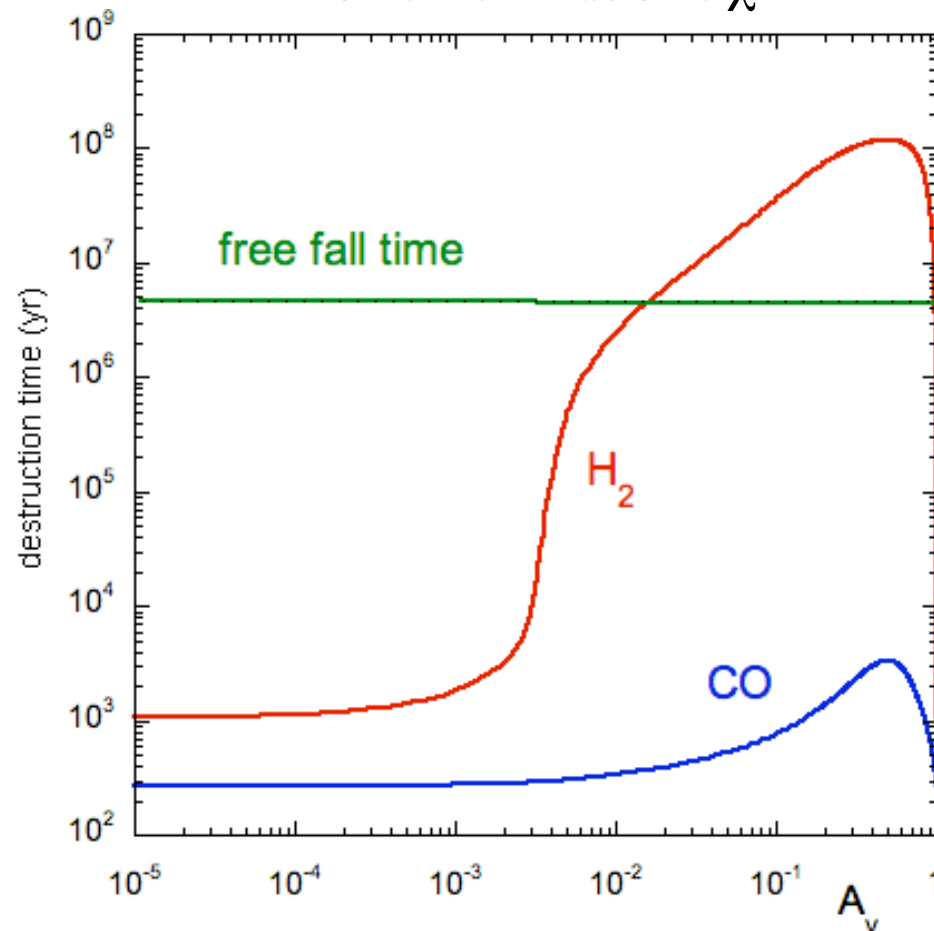
Random variation of reaction rate coefficients by a factor of 10



PDR models

Stationarity & time dependence $t_{dest}(X) = \frac{n(X)}{\text{destruction rate in cm}^{-3} \text{ s}^{-1}}$

$n_H = 100 \text{ cm}^{-3}$, $\zeta = 5 \cdot 10^{-17} \text{ s}^{-1}$,
2 side illumination : $\chi = 1$



Compare chemical times to dynamical times

$$t_{\text{free fall}} = \sqrt{\frac{3\pi}{32G\rho(0)}} = \frac{4.3 \times 10^7}{\sqrt{n_H(0)}} \text{ yrs}$$

- at the edge, destruction by photons
- At higher A_v , destruction by chemical processes, cosmic rays

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H₂ Formation

“Observational” determination of H₂ formation rate in diffuse clouds

- Jura (1975) Copernicus $R = 3 \times 10^{-17} \text{ cm}^3 \text{ s}^{-1}$
- Gry et al. (2002) FUSE $R = 3.1 - 4.5 \times 10^{-17} \text{ cm}^3 \text{ s}^{-1}$

H₂ formation rate in the Meudon PDR code

- fixed formation rate
- calculated from grain properties
- calculated from rate equations, moment equations (in progress)

Grain size distribution : $dn(a) \propto a^\alpha da$ with $\alpha = -3.5$ Mathis et al. (1983)

$$\frac{dn(H_2)}{dt} = \frac{1}{2} s v_H n(H) \langle \sigma_g \cdot n_g \rangle$$

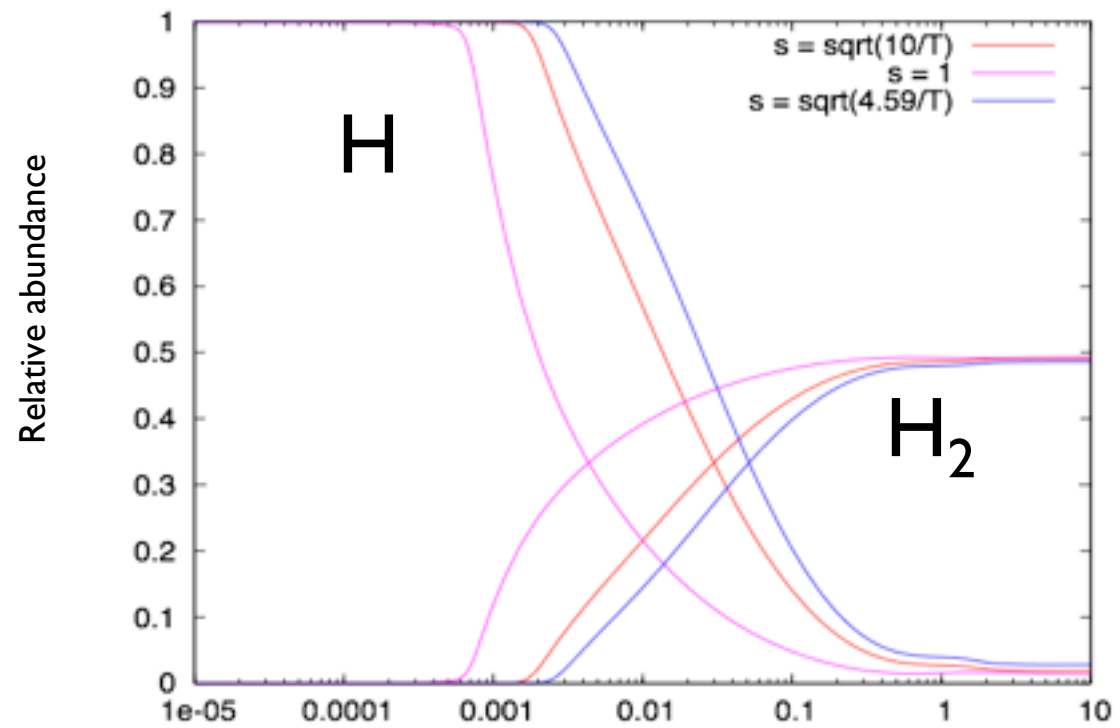
$$\langle \sigma_g \cdot n_g \rangle = \frac{3}{4} \frac{1.4 m_H G}{\rho_g} \frac{1}{\sqrt{a_{min} \cdot a_{max}}} n_H$$

$$\left. \frac{dn(H_2)}{dt} \right|_{form.gr.} = s \cdot 1.4 \times 10^{-17} \sqrt{T_K} n_H n(H)$$

H₂ Formation

Different choices for s :

- $s = \sqrt{(10/T)}$: R independent de T
- $s = 1$
- $s = \sqrt{(4.59/T)}$ gives $R = 3 \times 10^{-17} \text{ cm}^3 \text{ s}^{-1}$
- ...



H₂ Formation

case of small grains

If the mean number of H atoms is about 1, the rate equations formalism does not apply anymore.

Master equation approach: Biham et al. 2001, ApJ 553, 595

$$\begin{aligned} \frac{dP_H(N_H)}{dt} = & F_H [P_H(N_H - 1) - P_H(N_H)] \\ & + W_H [(N_H + 1)P_H(N_H + 1) - N_H P_H(N_H)] \\ & + A_H [(N_H + 2)(N_H + 1)P_H(N_H + 2) - N_H(N_H - 1)P_H(N_H)] \end{aligned}$$

with :

F_H : flux of H atoms impinging on the surface

W_H : desorption rate of H atoms : $W_H = \nu \exp(-E_1/kT_g)$

A_H : scanning of the surface by H atoms: $A_H = a/S$, where S is the number of adsorption sites on the grain ($S=4\pi r^2 s$) and $a = \nu \exp(-E_0/kT_g)$

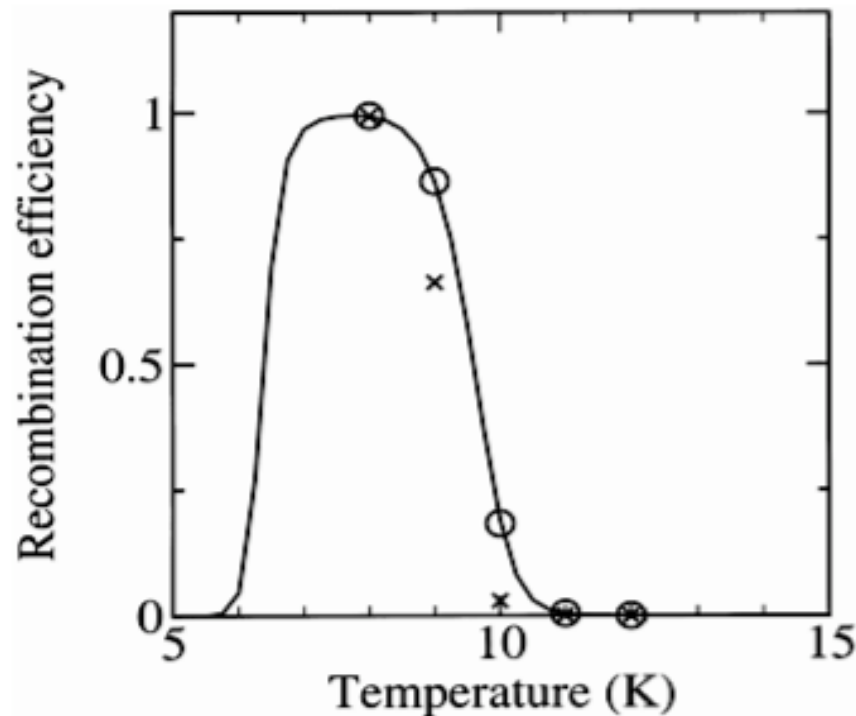
H₂ formation rate:

$$R_{H_2}^{grain} = A_H (\langle N_H^2 \rangle - \langle N_H \rangle^2)$$

$\nu (s^{-1})$	10^{12}
$s (cm^{-2})$	$5 \cdot 10^{13}$
$E_0 (K)$	510.6
$E_1 (K)$	658

H₂ Formation

Biham et al. 2001



- × master equation, $d = 0.01 \mu\text{m}$
- master equation, $d = 0.1 \mu\text{m}$
- rate equation

H₂ formation efficient only in a narrow window of grain temperatures

$$T_0 < T_g < T_1$$

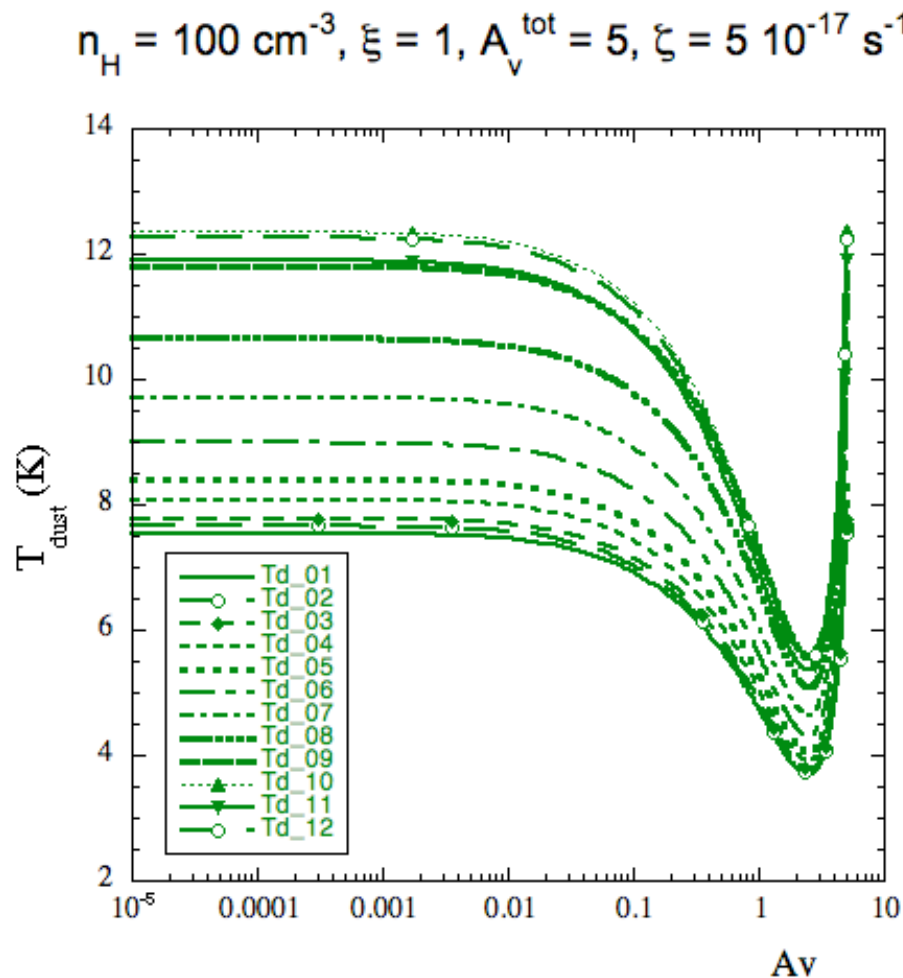
$$T_0 = \frac{E_0}{k_B \ln(\nu S/F)}$$

$$T_1 = \frac{2E_1 - E_0}{k_B \ln(\nu S/F)}$$

Determination of T_g essential

Determination of grain temperatures

- Observational information ?
- Thermal balance : approximate formula of Tielens & Hollenbach with simple assumptions on the grain absorption and emission coefficients (used in the Meudon PDR code)
- detailed calculations (cf Cuppen, Herbst, ...)



Case of the MRN size distribution

$$dn_g \approx a^{-3.5}$$

$$a_{\text{min}} = 3 \cdot 10^{-7} \text{ cm}$$

$$a_{\text{max}} = 3 \cdot 10^{-5} \text{ cm}$$

The temperatures at the edge have the right order of magnitude for H_2 formation.

Meudon PDR code

Conclusions

Detailed UV radiative transfer

Introduction of the impinging radiation field (from 912 Å to IR)

Detailed microscopic physics

Versatile chemistry; possibility of state to state chemistry

Source and input data downloadable from <http://aristote.obspm.fr/MIS>

Inclusion in the Virtual Observatory

Plane parallel geometry

Steady state

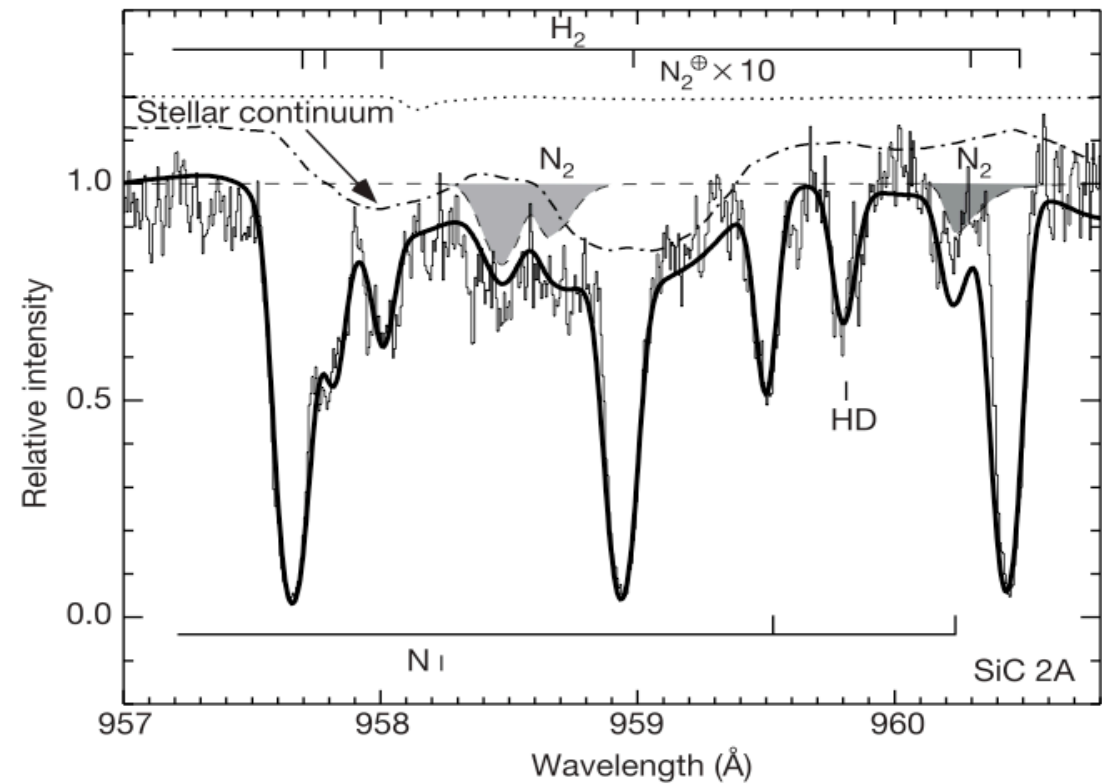
Extension to X radiation

Grain physics and chemistry should be improved (in particular H₂ formation)

Computing time becomes excessive when “exact” transfer is performed (but essentially needed for diffuse environments)

Meudon PDR code

Knauth et al., Nature, 429, 2004



Exemple : Detection of N_2

- HD 124314 (Knauth et al. 2004)
- 20 Aquilae (Knauth et al.)