

# MODULO: models of proto galaxies

Chemical evolution with a  
multizone multiphase model

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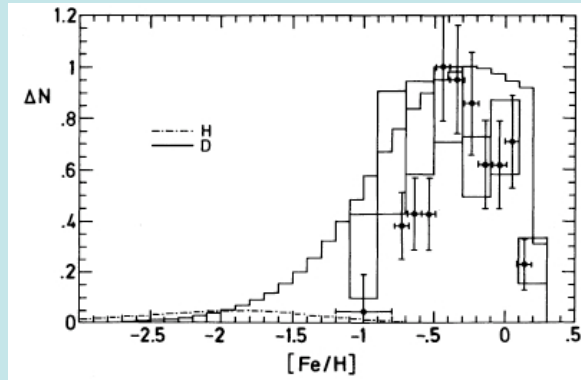
# The model

Designed to study galaxies at present-time...and in their recent past:

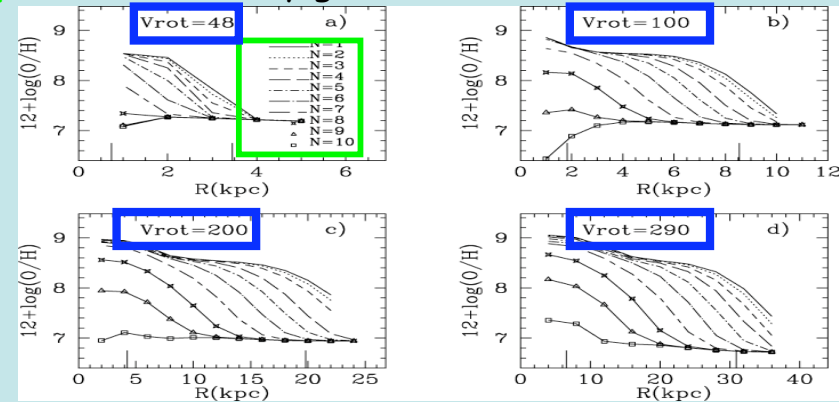
- Solar neighbourhood (e.g., Ferrini et al. 1992)
- Galactic halo, disk (thin and thick) and bulge (e.g., metallicity gradients, Ferrini et al. 1994, Magrini et al. 2009a)
- External galaxies (elliptical, spirals, irregulars, e.g. Molla' & Diaz 2005, Magrini et al. 2007)

# Some results:

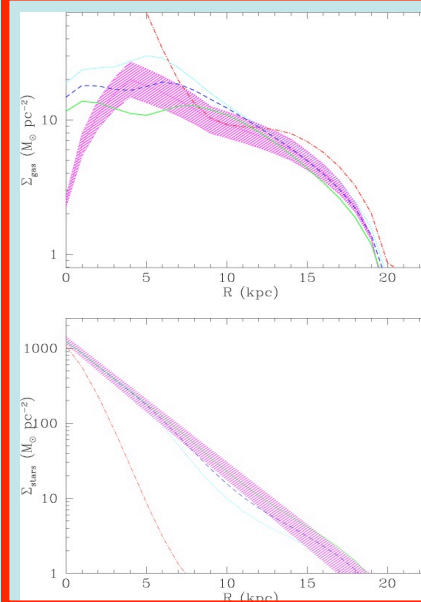
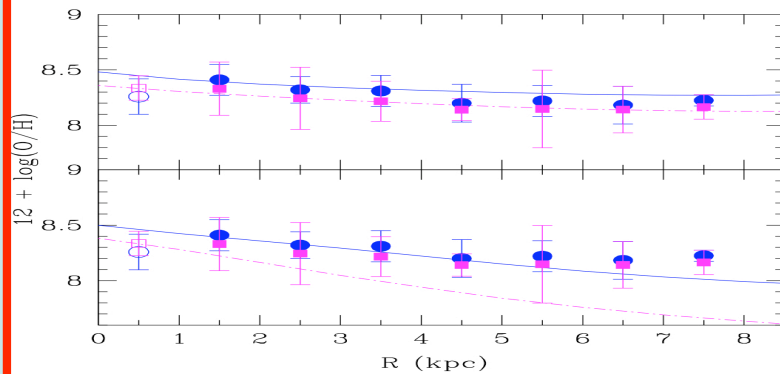
The solution of the G-dwarfs Problem (e.g. Ferrini et al. 1992)



The link between mass, morphological Type and metallicity gradients (Molla' & Diaz 2005)



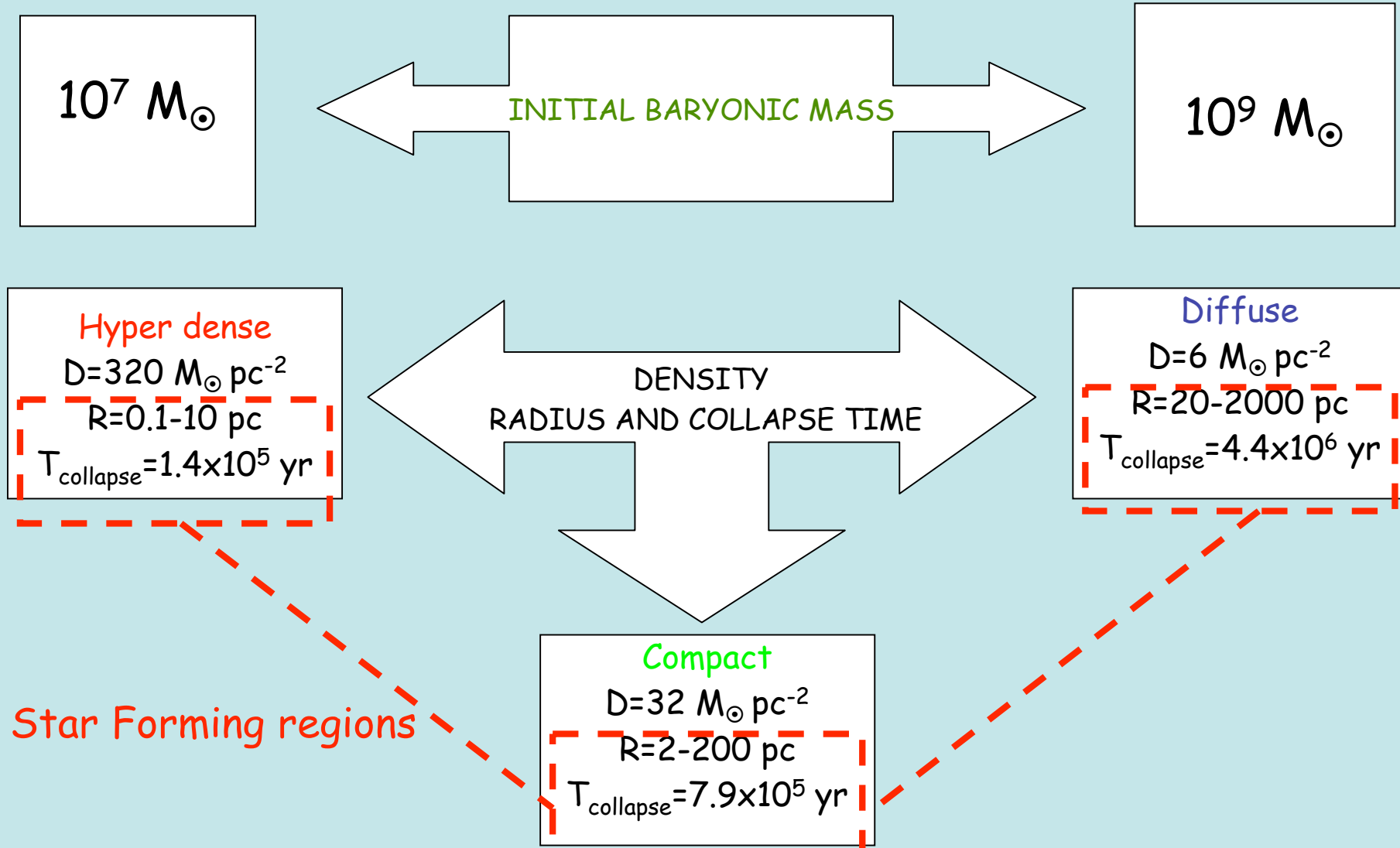
The time evolution of the metallicity gradient of M33 (Magrini et al. 2009b)



The radial distribution of the different baryonic components in the MW (e.g. Magrini et al. 2009a)

# Characteristics of proto-galaxies

(Hunt & Hirashita):



# From HII regions to galaxies

- Given the characteristic of the star formation regions (radius and collapse time), we consider larger regions containing the initially available mass of gas ( $10^7$  and  $10^9 M_{\odot}$ ) and the given surface densities,  $\Sigma = 320, 32, 6 M_{\odot} \text{pc}^{-2}$

$$R = \sqrt{M / \pi \Sigma}$$

and a collapse time corresponding to the free fall time of the host galaxy

$$\tau = \sqrt{3\pi / 32 G \rho} \text{ where } \rho = M / (4/3 \pi R^3)$$

# Size of the regions/galaxies

Hyper dense	Compact	Diffuse
$R = \sqrt{(10^7 / \pi 320)} =$ 100 pc $T = 5.2 \cdot 10^6$ yr	$R = \sqrt{(10^7 / \pi 32)} =$ 320 pc $T = 3 \cdot 10^7$ yr	$R = \sqrt{(10^7 / \pi 6)} =$ 730 pc $T = 10^8$ yr
$R = \sqrt{(10^9 / \pi 320)} =$ 1 kpc $T = 1.5 \cdot 10^7$ yr	$R = \sqrt{(10^9 / \pi 32)} =$ 3.2 kpc $T = 9 \cdot 10^7$ yr	$R = \sqrt{(10^9 / \pi 6)} =$ 7.3 kpc $T = 8 \cdot 10^8$ yr

## Initial conditions: SF efficiency

- We consider a star formation efficiency of 0.1 meaning that after  $\approx$  a dynamical time 10% of the total mass is converted into stars
- Thus the SFR is inversely proportional to the collapse time, being higher in the more compact objects
- The initial gas mass infall onto an inner disk where clouds and stars are formed

## The equations of the model:

### Star formation

$$\frac{ds_{1,D}}{dt} = H_{1,D}c_D^2 + a_{1,D}c_Ds_{2,D} - D_{1,D},$$

$$\frac{ds_{2,D}}{dt} = H_{2,D}c_D^2 + a_{2,D}c_Ds_{2,D} - D_{2,D},$$

$$\frac{dg_D}{dt} = -\mu_Dg_D^n + a'_Dc_Ds_{2,D} + H'_Dc_D^2 + W_D + fg_H, \quad \text{Infall}$$

### Cloud formation

$$\frac{dc_D}{dt} = \mu_Dg_D^n - (a_{1,D} + a_{2,D} + a'_D)c_Ds_{2,D} - (H_{1,D} + H_{2,D} + H'_D)c_D^2$$

$$\frac{dr_D}{dt} = D_{1,D} + D_{2,D} - W_D$$



# Atomic gas, molecular clouds and stars:

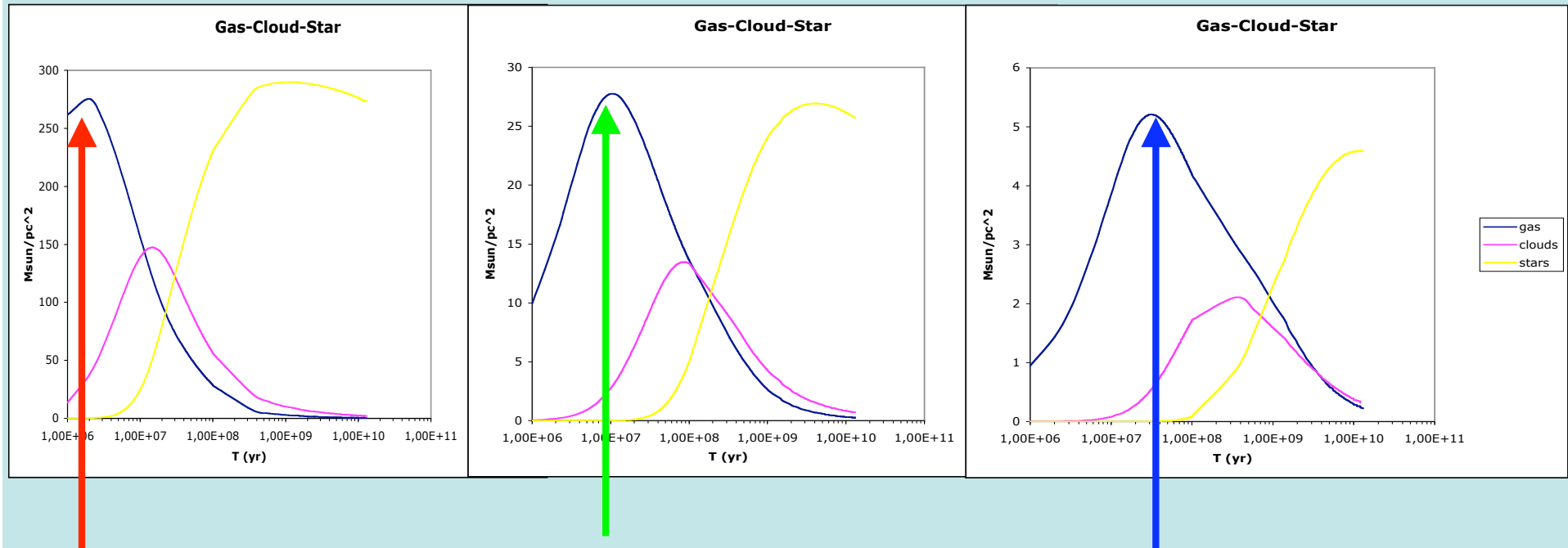
Hyper dense

-->

compact

-->

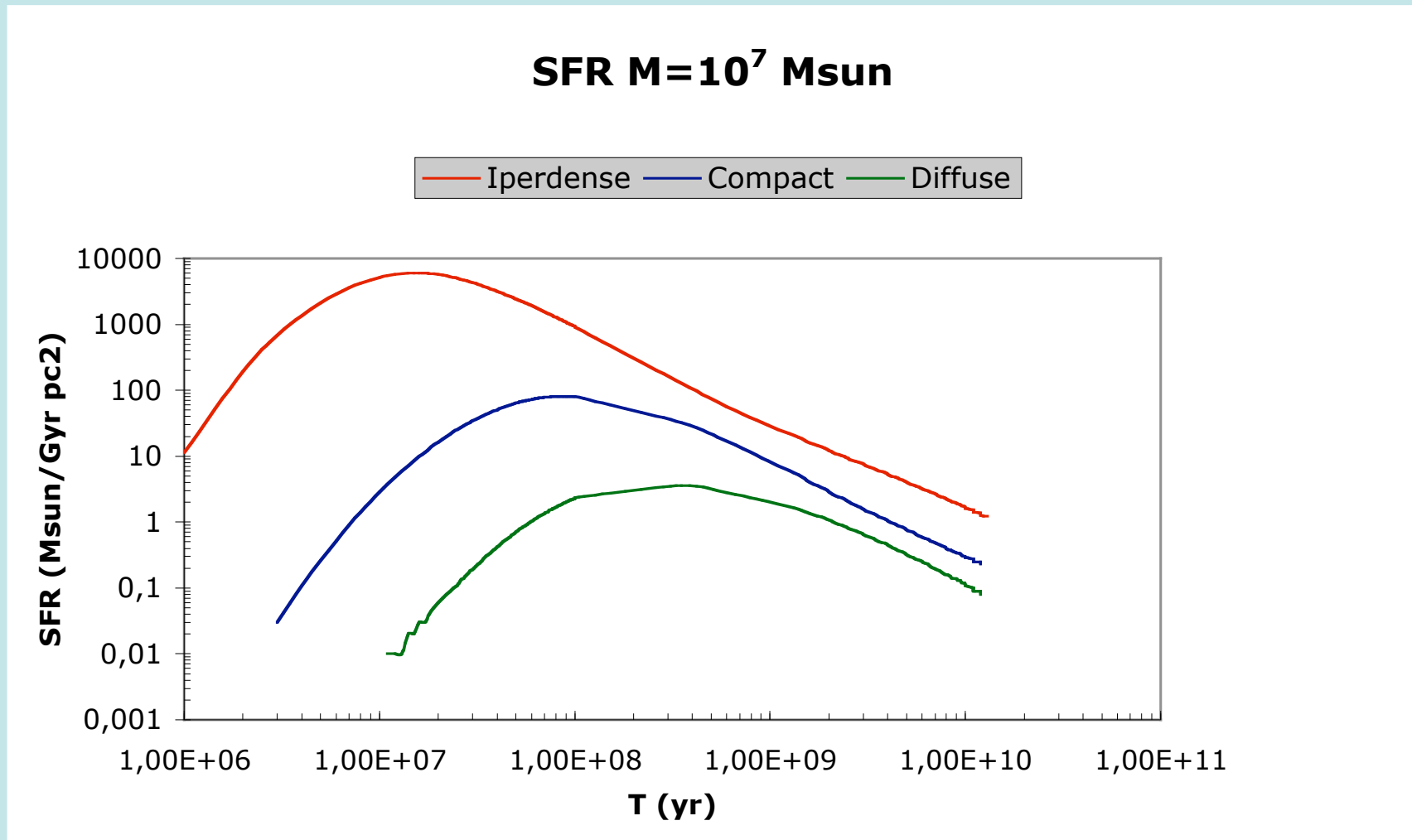
diffuse



Peak of the gas falling from the 'halo':  $2 \times 10^6$ ,  $1 \times 10^7$ ,  $3 \times 10^7$  yr

- Infall time ( $\propto$  collapse free fall time)
- Star and cloud formation efficiencies

# Star formation histories:

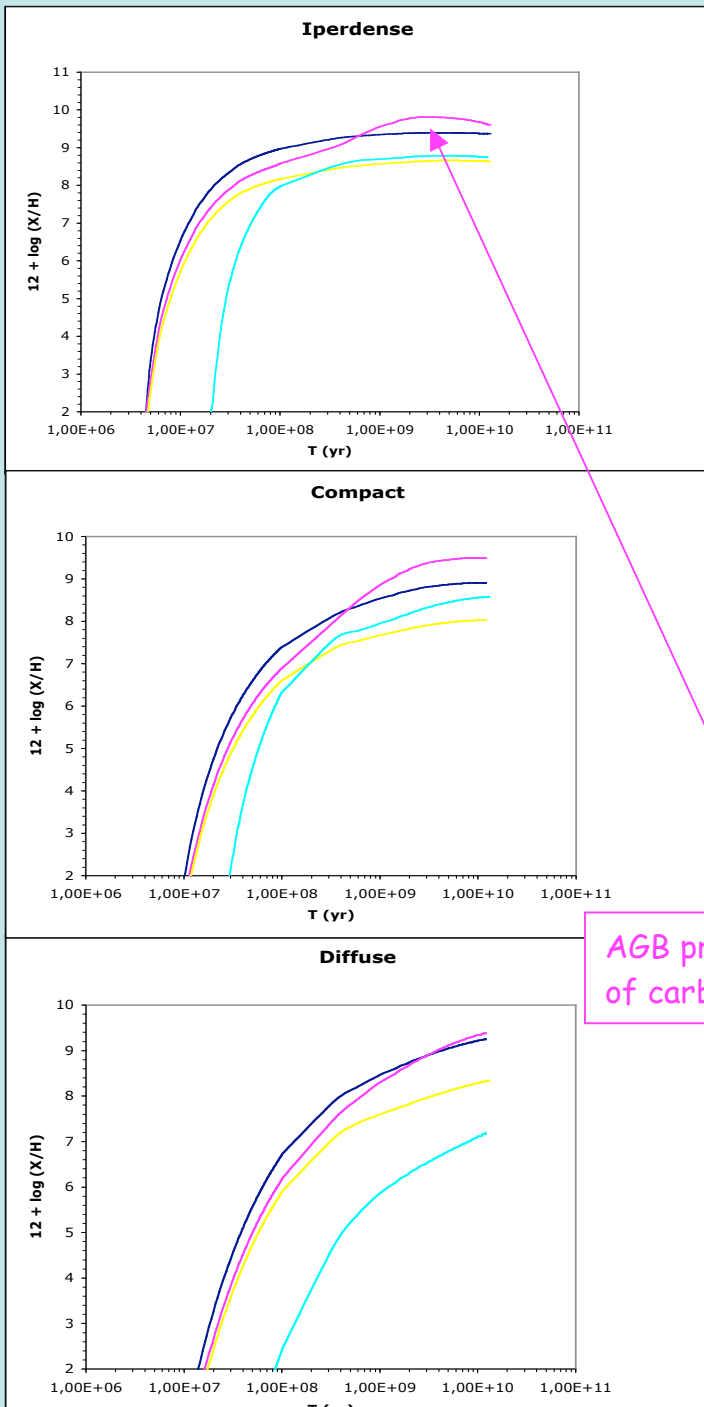


The peak of the SF depends essentially on the collapse time

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# ISM chemical abundances



- Chemical abundances of major constituents of molecules
  - $\alpha$ -elements (produced mainly by massive stars through SNII)  
oxygen, carbon, silicon
  - elements produced by intermediate mass stars through AGB phase: nitrogen (but also carbon)

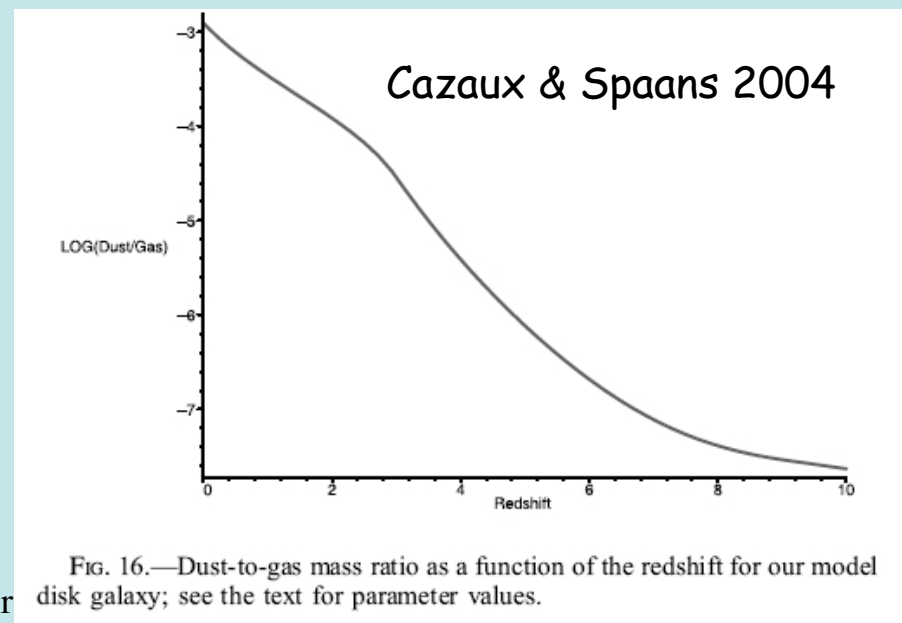
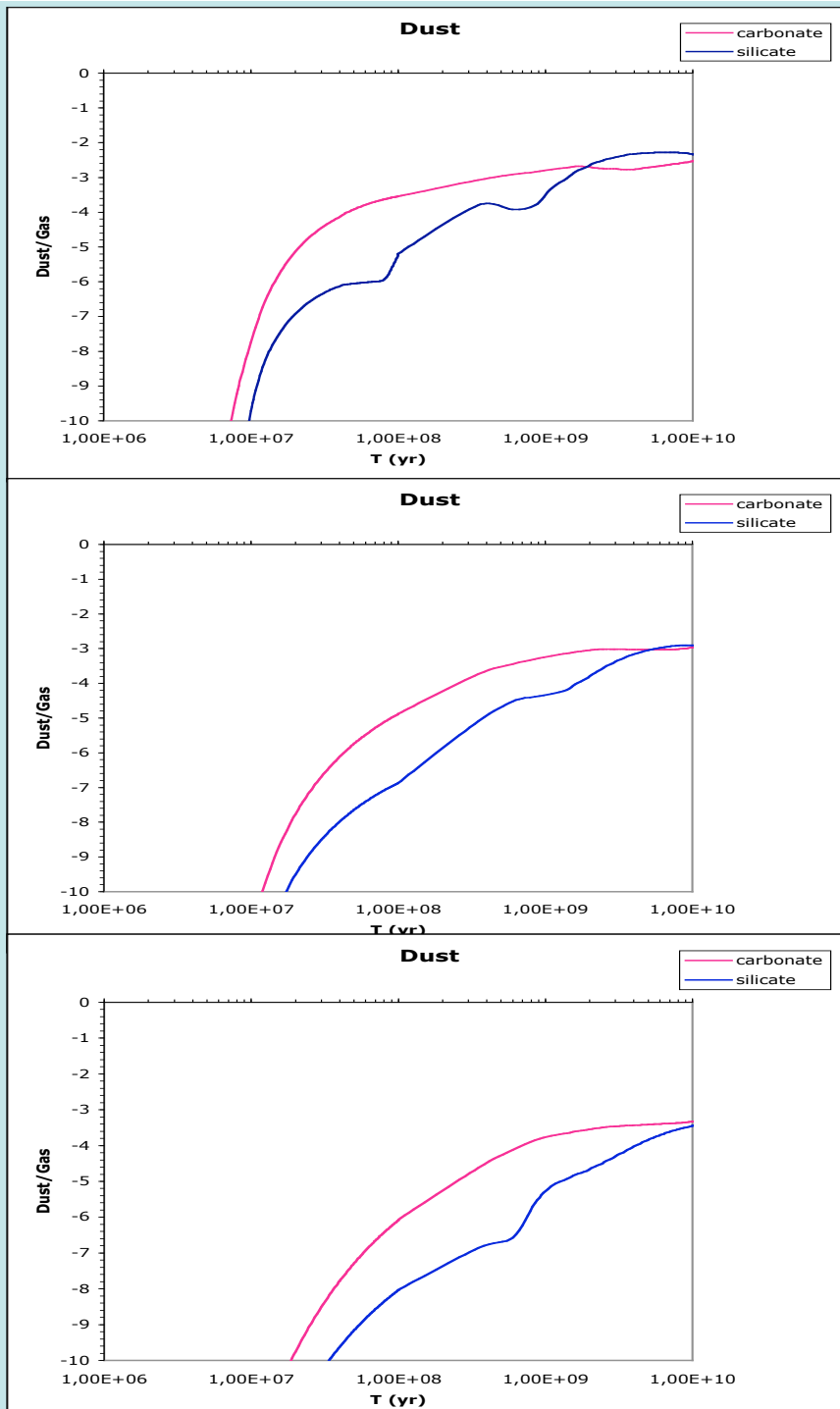
# The evolution of dust

Dust (carbonate and silicate)  
produced by

→ AGB stars (Zhukobska 2008)

→ Supernovae (Bianchi & Schneider 2007)

but also destroyed by SNe



): Fir

FIG. 16.—Dust-to-gas mass ratio as a function of the redshift for our model disk galaxy; see the text for parameter values.

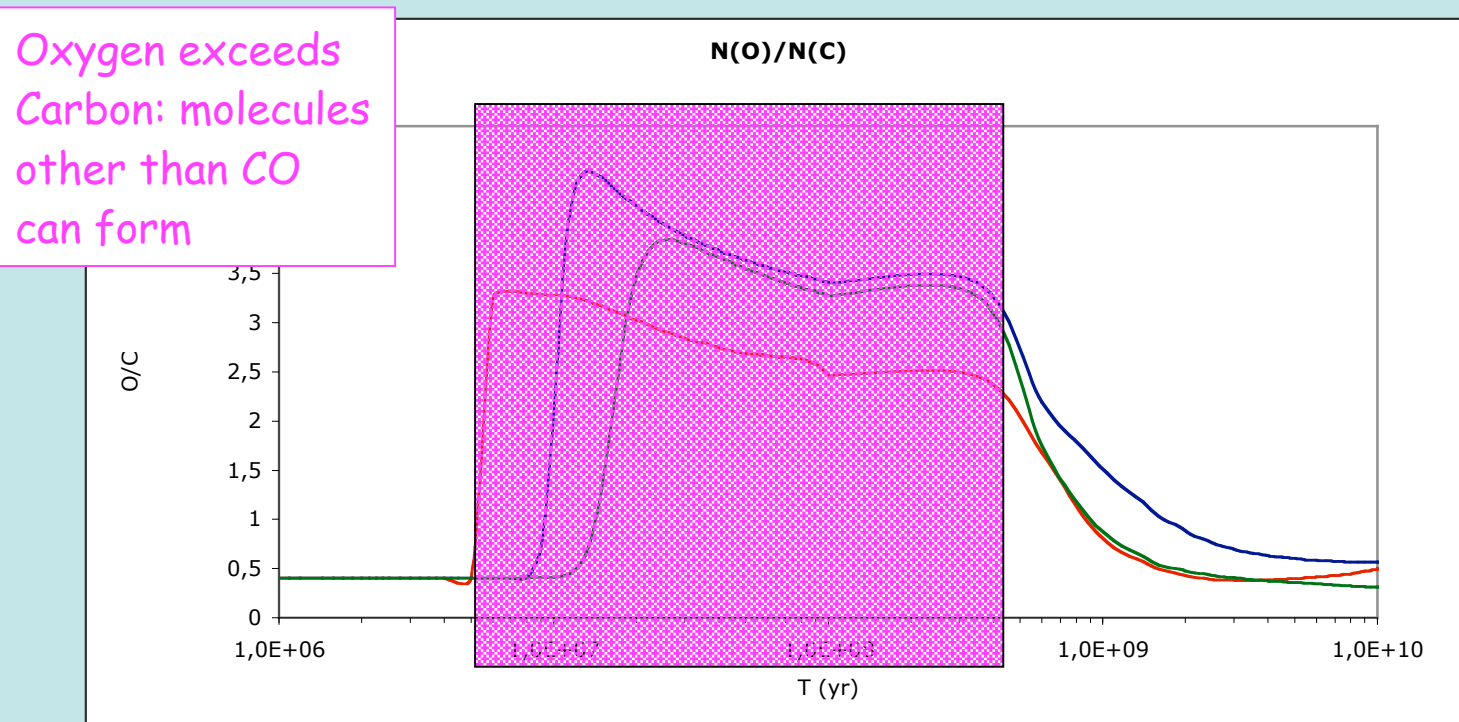
# Detected ISM molecules

H <sub>2</sub>	HD	The most abundant molecules			C <sub>2</sub> H	*C <sub>3</sub>		
CH	CH <sub>2</sub>				CH <sub>4</sub>	C <sub>4</sub>		
CH <sub>3</sub>	C <sub>2</sub> H <sub>2</sub>				C <sub>5</sub>	*C <sub>2</sub> H <sub>4</sub>	C <sub>5</sub> H	
c-C <sub>3</sub> H <sub>2</sub>	H <sub>2</sub> C <sub>3</sub>				*HC <sub>6</sub> H	H <sub>2</sub> C <sub>6</sub>		
H <sub>2</sub> C <sub>4</sub> (lin)	*HC <sub>4</sub> H	CH <sub>3</sub> C <sub>2</sub> H	C <sub>6</sub> H					
*C <sub>7</sub> H	CH <sub>3</sub> C <sub>4</sub> H	C <sub>8</sub> H	*C <sub>6</sub> H <sub>6</sub>					
OH	CO	CO+	H <sub>2</sub> O	HCO	HCO+			
HOC+	C <sub>2</sub> O	CO <sub>2</sub>	H <sub>3</sub> O+	HOCO+	H <sub>2</sub> CO			
C <sub>3</sub> O	CH <sub>2</sub> CO	HCOOH	H <sub>2</sub> COH+	CH <sub>3</sub> OH	CH <sub>2</sub> CHO			
CH <sub>2</sub> CHOH	CH <sub>2</sub> CHCHO		HC <sub>2</sub> CHO	C <sub>5</sub> O	CH <sub>3</sub> CHO	c-C <sub>2</sub> H <sub>4</sub> O		
CH <sub>3</sub> OCHO	CH <sub>2</sub> OHCHO		CH <sub>3</sub> COOH	CH <sub>3</sub> OCH <sub>3</sub>	CH <sub>3</sub> CH <sub>2</sub> OH	CH <sub>3</sub> CH <sub>2</sub> CHO		
(CH <sub>3</sub> ) <sub>2</sub> CO	HOCH <sub>2</sub> CH <sub>2</sub> OH		C <sub>2</sub> H <sub>5</sub> OCH <sub>3</sub>	(CH <sub>2</sub> OH) <sub>2</sub> CO				
NH	CN	N <sub>2</sub>	NH <sub>2</sub>	HCN	HNC			
N <sub>2</sub> H <sup>+</sup>	NH <sub>3</sub>	HCNH <sup>+</sup>	H <sub>2</sub> CN	HCCN	C <sub>3</sub> N			
CH <sub>2</sub> CN	CH <sub>2</sub> NH	HC <sub>2</sub> CN	HC <sub>2</sub> NC	NH <sub>2</sub> CN	C <sub>3</sub> NH			
CH <sub>3</sub> CN	CH <sub>3</sub> NC	HC <sub>3</sub> NH <sup>+</sup>	*HC <sub>4</sub> N	C <sub>5</sub> N	CH <sub>3</sub> NH <sub>2</sub>			
CH <sub>2</sub> CHCN	HC <sub>5</sub> N	CH <sub>3</sub> C <sub>3</sub> N	CH <sub>3</sub> CH <sub>2</sub> CN	HC <sub>7</sub> N	CH <sub>3</sub> C <sub>5</sub> N?	HC <sub>9</sub> N	HC <sub>11</sub> N	
NO	HNO	N <sub>2</sub> O	HNCO	NH <sub>2</sub> CHO				
SH	CS	SO	SO+	NS	SiH			
*SiC	SiN	SiO	SiS	HCl	*NaCl			
*AlCl	*KCl	HF	*AlF	*CP	PN			
H <sub>2</sub> S	C <sub>2</sub> S	SO <sub>2</sub>	OCS	HCS+	c-SiC <sub>2</sub>			
*SiCN	*SiNC	*NaCN	*MgCN	*MgNC	*AlNC			
H <sub>2</sub> CS	HNCS	C <sub>3</sub> S	c-SiC <sub>3</sub>	*SiH <sub>4</sub>	*SiC <sub>4</sub>			
CH <sub>3</sub> SH	C <sub>5</sub> S	FeO						

Courtesy of V. Casasola

# The formation of molecules

- We consider a very simple chemistry to form molecules:  $\text{CO}$ ,  $\text{H}_2\text{O}$ ,  $\text{SiO}$ ,  $\text{HCN}$
- We consider  $\text{CO}$  as the most probable molecule to form, thus only the excess oxygen, (O-C), can form the other molecules, e.g.,  $\text{H}_2\text{O}$  and  $\text{SiO}$ .

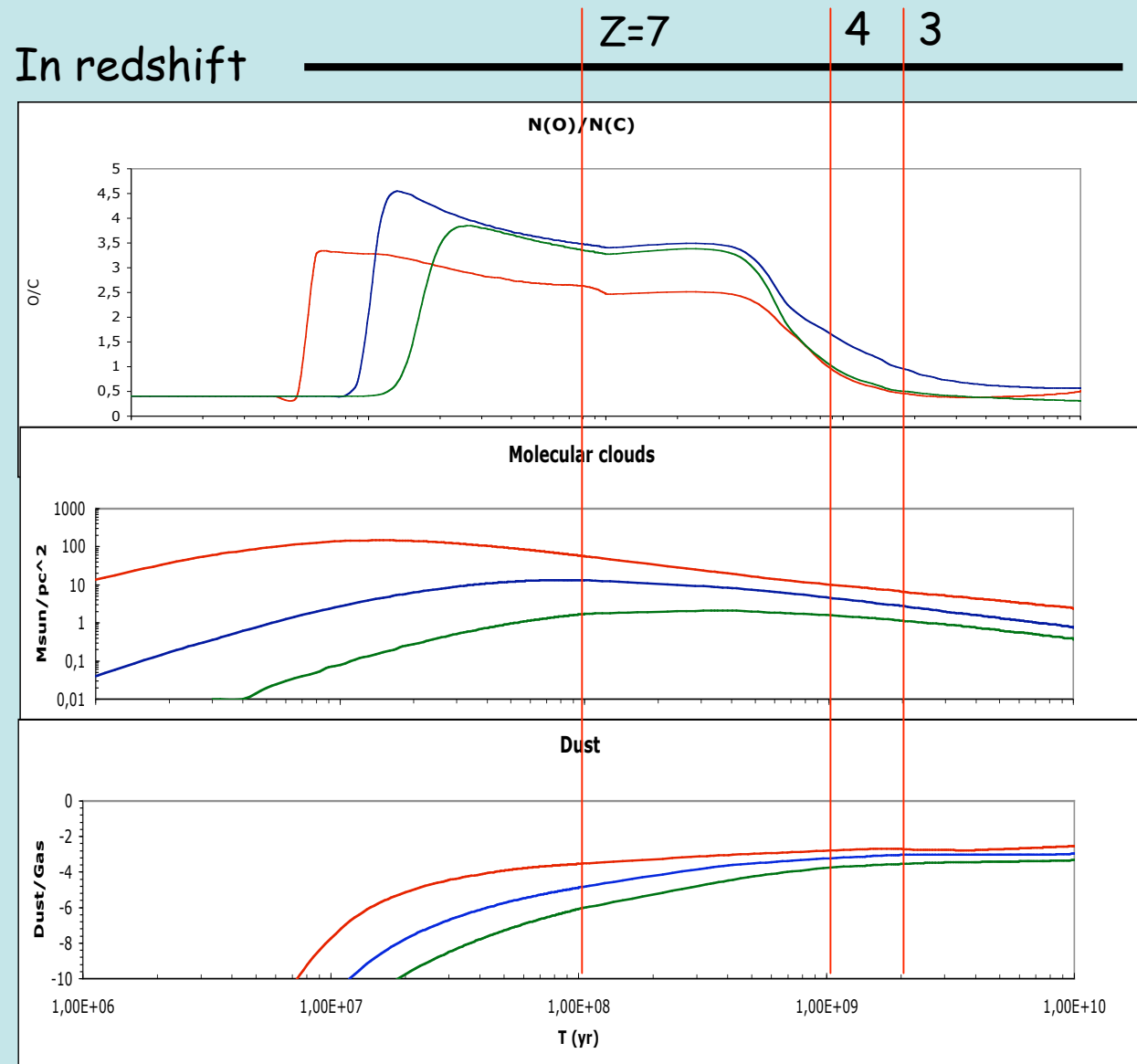


# What we need to form a molecule different than CO:

1.  $N(O) > N(C)$

2. Presence of  
Molecular clouds

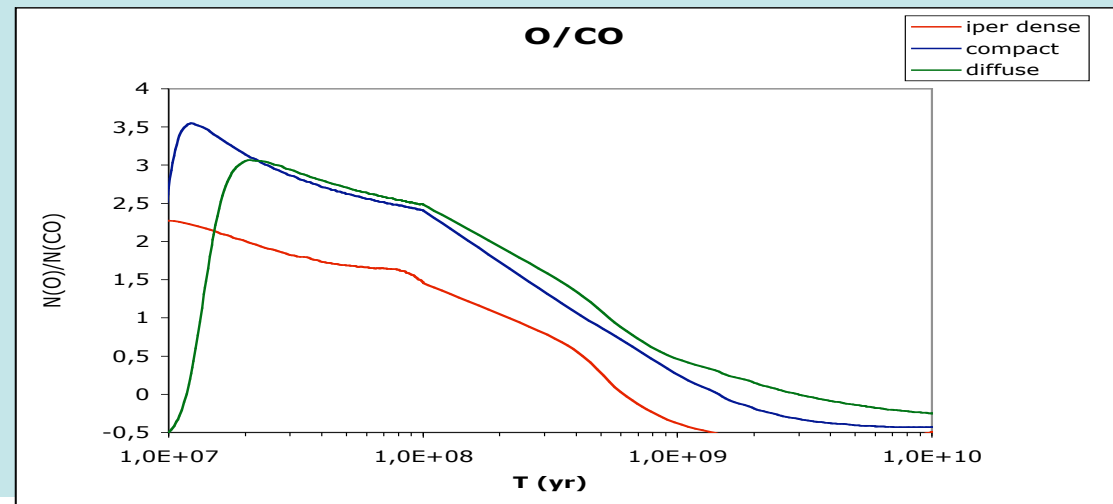
3. Dust to allow the  
formation of molecules  
on their surface



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# Comparison with a model of an elliptical galaxy:



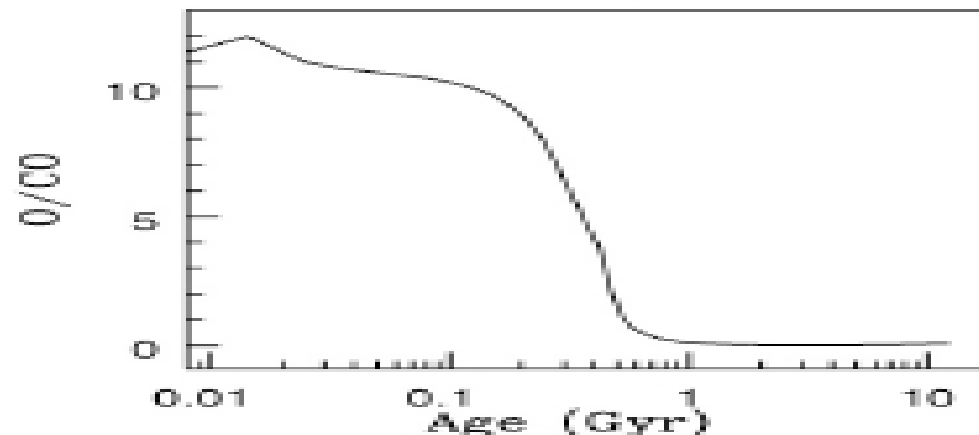
$z(\text{form}) = 7.5$

7 6 5 4 3 2 1

$z(\text{form}) = 10$

7 6 5 4 3 2 1

redshift (QSO)

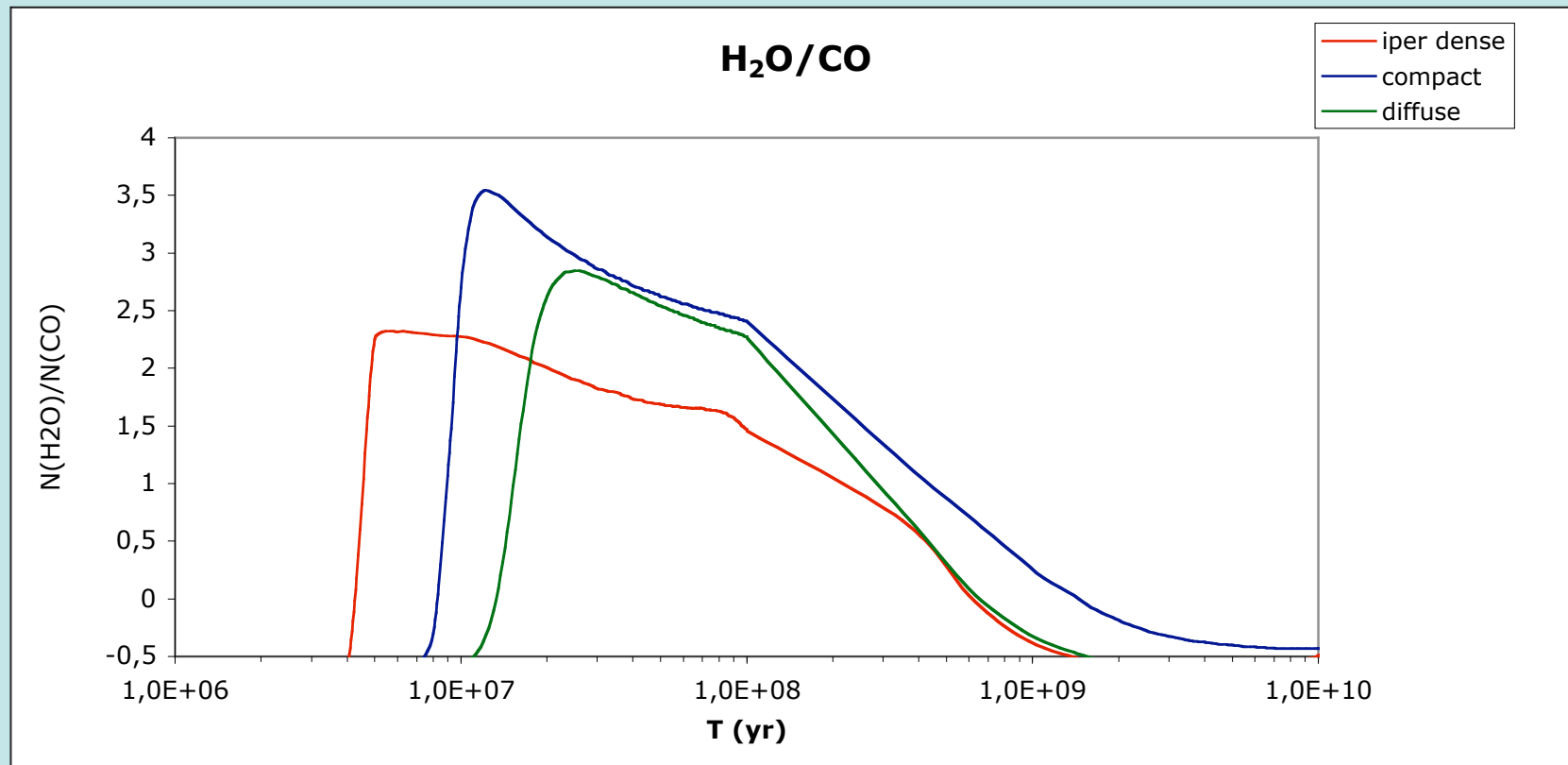


Matteucci & Pipino



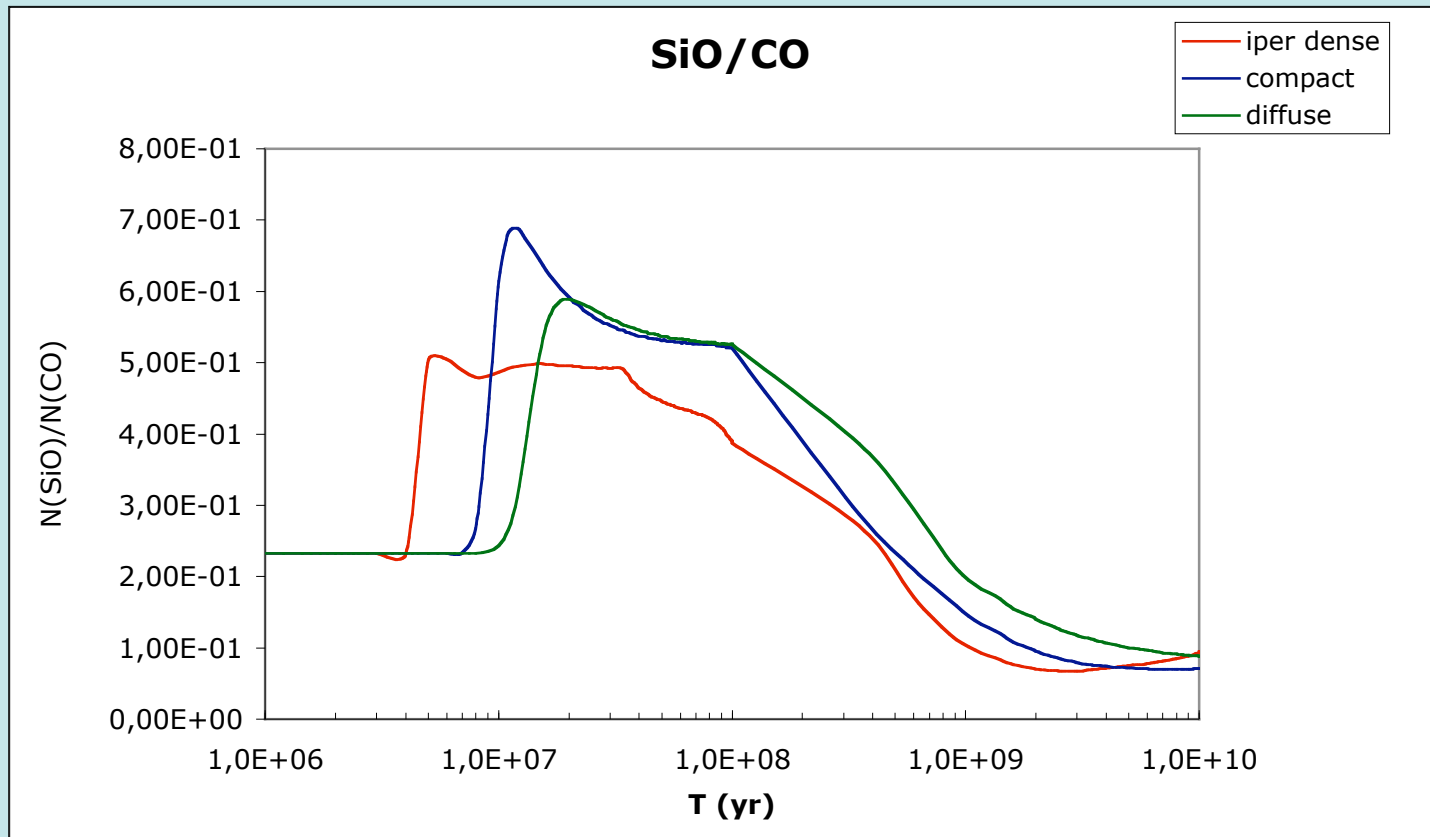
# The formation of water:

Assuming that all remaining oxygen after the formation of CO molecules would form  $H_2O \rightarrow n(H_2O) > n(CO)$  at  $\sim 0.01-0.1$  Gyr



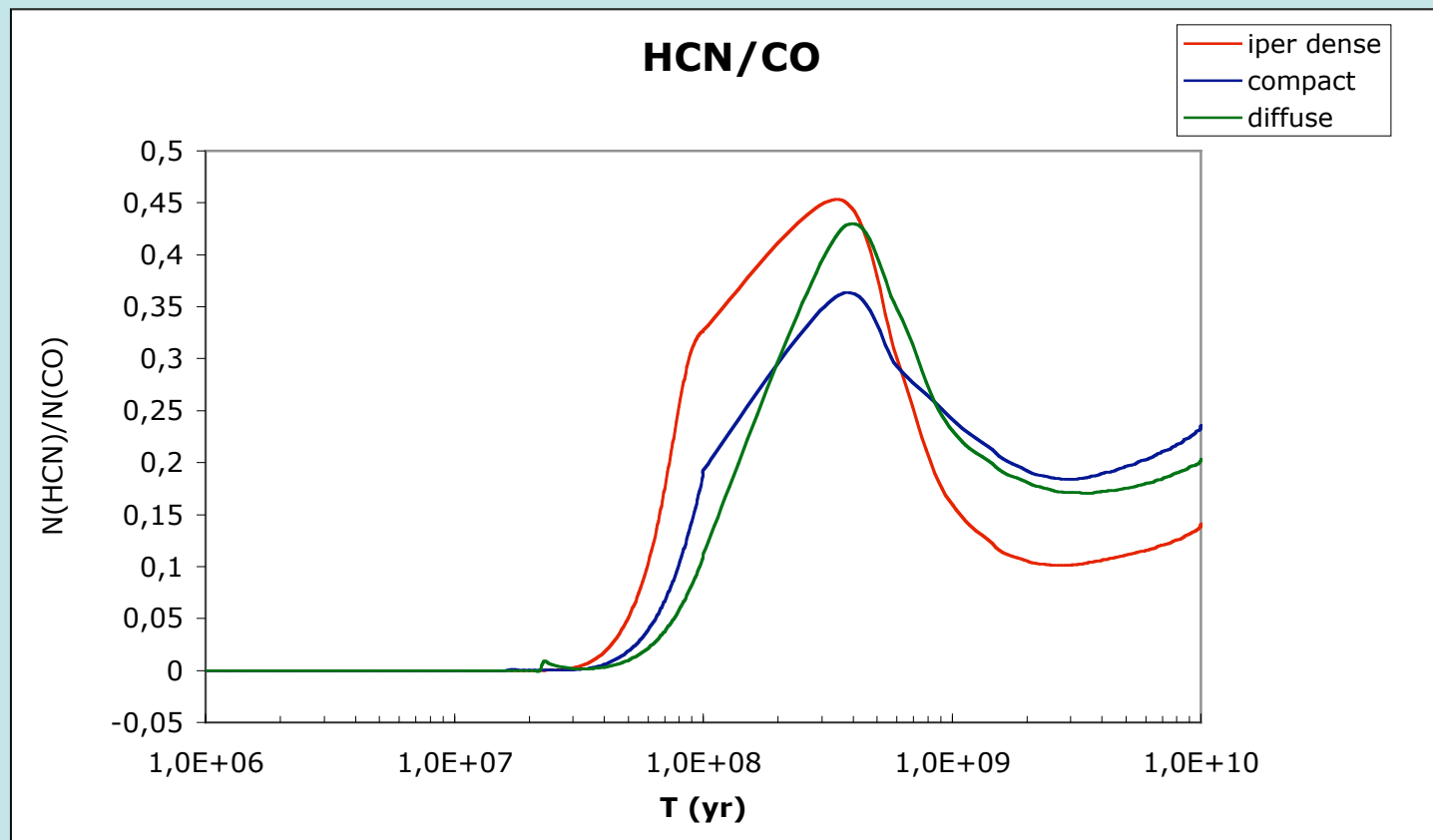
# The formation of SiO molecule:

Considering the minimum between  $N(O-C)$  and  $N(Si)$  respect to  $N(C)$  we compute the time evolution of the ratio  $N(SiO)/N(CO)$  --> due to the lower abundance of Si respect to C,  $N(SiO) \ll N(CO)$ , and having a maximum after 0.01-0.1 Gyr



# The formation of the HCN molecule:

When the production of N starts, we consider that all nitrogen is used to form HCN molecules, and thus CO is made with the remaining carbon.



# Conclusions:

Our chemical evolution model is able to follow the time evolution of:

A. atomic, molecular gas and stars surface densities

B. the ISM abundances of different chemical elements

C. the dust-to-gas ratio (total, carbonate, silicate)

We need a link to the processes of molecule formation

# Questions:

1. Is there a threshold in Dust-to-gas ratio and in molecular cloud density to form molecules?
1. What about the other molecules containing carbon like HCN? We would need something like PROBABILITY of molecule formation.....